CANADIAN JOURNAL OF CHEMISTRY. VOL. 47, 1969

- 3. R. E. HUFFMAN. Handbook of geophysics and space environments. *Edited by* S. Valle. McGraw-Hill Book Co., Inc., New York. 1965. DASA reaction rate handbook, DASA 1948. Defense Atomic Support Agency, Washington. 1967. 4. G. L. WEISSLER. Handbuch der Physik. Vol. XXI.
- *Edited by* S. Flugge. Springer-Verlag, Berlin. 1956. 5. J. A. R. SAMSON. Techniques of vacuum ultraviolet
- spectroscopy. John Wiley and Sons, Inc., New York. 1967.
- 6. J. R. MCNESBY and H. OKABE. Advan. Photochem. 3, 147 (1964). 7. A. C. G. MITCHELL and M. W. ZEMANSKY. Reso-
- nance radiation and excited atoms. Cambridge University Press, London. 1934.
- 8. R. M. GOODY. Atmospheric radiation. I. Theoretical basis. The Clarendon Press, Oxford. 1964.
- 9. A. J. BLAKE, J. H. CARVER, and G. N. HADDAD. J. Quant. Spectry. Radiative Transfer, 6, 451 (1966).
- 10. J. A. R. SAMSON and R. B. CAIRNS. J. Geophys. Res. 69. 4583 (1964).
- 11. V. D. MEVER, A. SKERBELE, and E. N. LASSETTRE. J. Chem. Phys. 43, 3769 (1965).
- 12. J. A. R. SAMSON. Advan. At. Mol. Phys. 2, 177 (1966).
- G. V. MARR. Photoionization processes in gases. Academic Press, Inc., New York. 1967.
 R. W. DITCHBURN and P. A. YOUNG. J. Atmospheric Terrest. Phys. 24, 127 (1962).
 B. A. THOMPSON, P. HARTECK, and R. R. REEVES. Conclume Dec. 69 (421) (1962).
- J. Geophys. Res. 68, 6431 (1963).
- 16. P. H. METZGER and G. R. COOK. J. Quant. Spectry. Radiative Transfer, 4, 107 (1964). 17. R. E. HUFFMAN, Y. TANAKA, and J. C. LARRABEE.
- Discussions Faraday Soc. 37, 154 (1964). 18. R. GOLDSTEIN and F. N. MASTRUP. J. Opt. Soc.
- Am. 56, 765 (1966).
- 19. J. O. SULLIVAN and P. WARNECK. J. Chem. Phys.
- 46, 953 (1967).
 20. R. A. YOUNG and G. BLACK. J. Chem. Phys. 47, 2311 (1967).
- 21. J. S. EVANS and C. J. S. Technical Report R-92 (1961). SCHEXNAYDER, NASA
- R. D. HUDSON, V. L. CARTER, and J. A. STEIN. J. Geophys. Res. 71, 2295 (1966).
 R. E. HUFFMAN, J. C. LARRABEE, and Y. TANAKA.
- J. Chem. Phys. 40, 356 (1964).
- 24. G. R. COOK and P. H. METZGER. J. Chem. Phys. 41, 321 (1964).

- 25. F. M. MATSUNAGA and K. WATANABE. Sci. Light, Tokyo, **16**, 31 (1967).
- W. F. J. EVANS, D. M. HUNTEN, E. J. LLEWELLYN, 26. and A. VALLANCE JONES. J. Geophys. Res. 73, 2885 (1968).
- 27. R. P. WAYNE. Quart. J. Roy. Meteorol. Soc. 93, 69 (1967).
- 28. R. E. HUFFMAN, J. C. LARRABEE, and Y. TANAKA.
- J. Chem. Phys. 46, 2213 (1967).
 29. R. E. HUFFMAN, J. C. LARRABEE, and Y. TANAKA. J. Chem. Phys. 47, 4462 (1967).
- 30. F. ALBERTI, R. A. ASHBY, and A. E. DOUGLAS. Can. J. Phys. 46, 337 (1968).
- R. E. HUFFMAN, Y. TANAKA, and J. C. LARRABEE. J. Chem. Phys. 39, 910 (1963).
- 32. K. CODLING. Astrophys. J. 143, 552 (1966).
- 33. J. P. APPLETON and M. STEINBERG. J. Chem. Phys. 46, 1521 (1967).
- G. M. LAWRENCE, D. L. MICKEY, and K. DRESSLER. J. Chem. Phys. 48, 1989 (1968).
- 35. M. OGAWA and R. B. CAIRNS. Planetary Space Sci. 12, 656 (1964).
- 36. A. DALGARNO, R. J. W. HENRY, and A. L. STEWART. Planetary Space Sci. 12, 235 (1964), and references therein
- 37. R. B. CAIRNS and J. A. R. SAMSON. Phys. Rev. 139, A1403 (1965).
- 38. F. J. COMES, F. SPEIER, and A. ELZER. Z. Naturforsch. 23a, 114 (1968)
- 39. R. J. W. HENRY. Planetary Space Sci. 15, 1747 (1967).
- R. E. HUFFMAN, J. C. LARRABEE, and Y. TANAKA. Phys. Rev. Letters, 16, 1033 (1966).
 E. C. Y. INN and Y. TANAKA. J. Opt. Soc. Am. 43,
- 870 (1953).
- 42. E. VIGROUX. Ann. Phys. 8, 709 (1953)
- P. A. LEIGHTON. Photochemistry of air pollution. Academic Press, Inc., New York. 1961, p. 48.
 E. VIGROUX. Ann. Phys. 2, 209 (1967).
- 45. W. B. DEMORE and O. RAPER. J. Chem. Phys. 44, 1780 (1966).
- 46. L. A. HALL, W. Schweizer, and H. E. HINTEREGGER. J. Geophys. Res. 70, 105 (1965).
- 47. J. A. R. SAMSON and R. B. CAIRNS. J. Opt. Soc. Am. 55, 1035 (1965).
- 48. R. E. HUFFMAN and J. C. LARRABEE. J. Geophys. Res. 73, 7419 (1968). 49. R. P. WAYNE. Priva
- Private communication.
- 50. J. C. LARRABEE and R. E. HUFFMAN. Unpublished results.

Absorption in the spectral range of the Schumann–Runge bands

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Absorption of silicon lines by O_2 in the Schumann-Runge bands has been measured. The results lead to some conclusions about the nature of the absorption and about the general behavior of the spectrum.

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Introduction

Early measurements, presented in Fig. 1, of the absorption coefficient of oxygen in the SchumannRunge bands (1750–2000 Å) and in the Herzberg continuum ($\lambda < 2424$ Å) were mainly performed with light sources providing spectral emission

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Fig. 1. Absorption cross section of molecular oxygen σ_{0_2} between 1700 and 2430 Å. The uncertainty remaining in the Schumann-Runge bands after measurements extending over a period of more than 60 years appears clearly.

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Band	Line	$v_{o_2}(cm^{-1})$	vsi(cm ⁻¹)	$\Delta v(cm^{-1})$	$\overline{\Delta v}(cm^{-1})$	S Bethke (cm)	r_{j}	$\substack{\sigma_{\nu_0}(O_2)\\(cm^2)}$	$\sigma_v(Si)$ (cm ²)	$\frac{\sigma_v(Si)}{\Sigma\sigma_{\nu_0}(O_2)}$
5-0	7P 9R 9P	52 513.2 52 511.9 52 486.5 52 485.5	52 502.7	10.5 9.2 16.2 17.2	13.2	6.44×10^{-19}	$\left(\begin{array}{c} 0.06589\\ 0.07434\\ 0.06690\\ 0.06664\end{array}\right)$	$\left. \begin{array}{c} 4.24 \times 10^{-20} \\ 4.79 \times 10^{-20} \\ 4.31 \times 10^{-20} \\ 4.29 \times 10^{-20} \end{array} \right\}$	5.26×10^{-22}	2.99×10^{-3}
0-9	19P 21R	52 823.3 52 822.6	52 819.2	4.1 3.4	3.8	1.53×10^{-18}	$\left\{ \begin{array}{c} 0.01186 \\ 0.01236 \end{array} \right.$	2.89×10^{-20} 1.89 × 10^{-20}	(1.31×10^{-21}) 1.23×10^{-21}	2.57×10^{-2}
0-9	13P 15R	52 973.4	52 974.7	1.3	1.3	1.53×10^{-18}	$\left\{ \begin{array}{c} 0.05102 \\ 0.04198 \end{array} \right\}$	7.81×10^{-20} 6.42×10^{-20}	(1.79×10^{-20}) 1.71 × 10^{-20}	1.20×10^{-1}
7–0	25P 27R	53 141.1 53 135.6	53 139.1	2.0 3.5	2.5	3.15×10^{-18}	$\left\{\begin{array}{c} 0.00380\\ 0.00204\end{array}\right.$	1.20×10^{-20} 6.43 × 10^{-21}	8.28×10^{-22}	4.49×10^{-2}
0-1	19P 21R 23R	53 349.1 53 345.4 53 285.6 53 281.4	53 310.2	38.9 35.2 24.6 28.8	33.2	3.15×10 ^{-,18}	$ \left\{\begin{array}{c} 0.01886\\ 0.01236\\ 0.01180\\ 0.00722 \end{array}\right. $	$\begin{array}{c} 5.94 \times 10^{-20} \\ 3.89 \times 10^{-20} \\ 3.72 \times 10^{-20} \\ 2.27 \times 10^{-20} \end{array}$	1.50×10^{-22}	1.05×10^{-3}
7–0	19P 21R	53 349.1 53 345.4	53 337.8	11.3 7.6	9.8	3.15×10^{-18}	$\begin{cases} 0.01886 \\ 0.01236 \end{cases}$	5.94×10^{-20} 3.89 × 10^{-20}	(4.06×10^{-22}) 4.28×10^{-22}	4.35×10^{-3}
2-0	17P 19R 19P 21R	53 406.7 53 403.6 53 349.1 53 345.4	53 387.3	19.4 16.3 38.2 41.9	26.6	3.15×10^{-18}	$ \begin{array}{c} 0.02821 \\ 0.01986 \\ 0.01886 \\ 0.01236 \end{array} $	$\begin{array}{c} 8.89 \times 10^{-20} \\ 6.26 \times 10^{-20} \\ 5.94 \times 10^{-20} \\ 3.89 \times 10^{-20} \end{array}$	3.14×10^{-22}	1.26×10^{-3}
8-0	13P 15R 17R	54 000.4 53 995.3 53 953.5 53 947.8	53 981.9	18.5 13.4 28.4 34.1	22.5	5.98×10^{-18}	$\left\{\begin{array}{c} 0.05102\\ 0.04198\\ 0.03936\\ 0.02987\end{array}\right.$	$\begin{array}{c} 3.05 \times 10^{-19} \\ 2.51 \times 10^{-19} \\ 2.35 \times 10^{-19} \\ 1.79 \times 10^{-19} \\ \end{array}$	(1.19×10^{-21}) 1.11×10^{-21}	1.14×10^{-3}
8-0	13R	54 036.8	54 034.4	2.4	2.4	5.98×10^{-18}	0.05494	3.29×10^{-19}	(1.40×10^{-20}) 1.31×10^{-20}	3.98×10^{-2}
8-0	7P 9R 9P 11R	54 104.1 54 101.5 54 075.6 54 072.2	54 090.7	13.4 10.8 15.1 18.5	14.4	5.98×10 ⁻¹⁸	$\left\{\begin{array}{c} 0.06589\\ 0.07434\\ 0.06690\\ 0.06664\end{array}\right.$	$\begin{array}{c} 3.94 \times 10^{-19} \\ 4.45 \times 10^{-19} \\ 4.00 \times 10^{-19} \\ 3.99 \times 10^{-19} \end{array}$	(2.96×10^{-21}) 3.39 × 10^{-21}	2.07×10^{-3}
8-0	7P 9R	54 104.1 54 101.5	54 108.2	4.1 6.7	5.5	5.98×10^{-18}	$\left\{ \begin{matrix} 0.06589 \\ 0.07434 \end{matrix} \right.$	3.94×10^{-19} 4.45×10^{-19}	(6.95×10^{-21}) 7.71 × 10^{-21}	9.19×10^{-3}
8-0	5P 7R	54 126.7 54 124.7	54 128.0	1.3	2.4	5.98×10^{-18}	$\left\{ \begin{array}{c} 0.05637\\ 0.07530 \end{array} \right.$	3.37×10^{-19} 4.50×10^{-19}	(3.20×10^{-20}) 3.83×10^{-20}	4.87×10^{-2}
8-0	RI 1₽ ↓ 05	54 157.3 54 152.7	54 167.9	10.6 15.2	13.4	5.98×10^{-18}	$\left\{\begin{array}{c} 0.02726\\ 0.01363\\ 0.01363\\ 0.01110\\ 0.01110\\ 0.01110\\ 0.01110\\ 0.01110\\ 0.01110\\ 0.01110\\ 0.01110\\ 0.01110\\ 0.01110\\ 0.010\\ 0.010\\ 0.010\\ 0.010\\ 0.010\\ 0.010\\ 0.000\\ 0$	1.63×10^{-19} 8.15 × 10^{-20} 2.05 × 10^{-19}	(1.71×10^{-21})	3.01×10^{-3}
0-6	23P	54 158.0		6.9		9.49×10^{-18}	0.00692	6.57×10^{-20}	01 × C0.1	

TABLE I Data

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	4	ACKER	MAN	ET AL	: Ał	350	RPTION I	ΝT	HE I	RAI	NGE OF	SCH	UMA	NN-J	RUNG	ΕB	ANDS			1	837	
$\frac{\sigma_{v}(Si)}{\Sigma\sigma_{v_{0}}(O_{2})}$	6.43×10^{-4}	1.88×10^{-2}	5.71×10^{-1}	1.49×10^{-3}	4.80×10^{-3}	3.02×10^{-1}	1.25×10^{-3}	1.18×10^{-2}	5.45×10^{-3}	1.25	1.98×10^{-2}			2.30×10^{-2}		5.54×10^{-1}	1.36×10^{-2}	7.76×10^{-2}		8.13×10^{-1}	1.04	
$\sigma_{v}(Si)$ (cm ²)	(5.50×10^{-22}) 5.12×10^{-22}	(2.06×10^{-21}) 2.11×10^{-21}	(6.57×10^{-21}) 7.54 $\times 10^{-21}$	(2.95×10^{-22}) 2.87×10^{-22}	(5.27×10^{-21}) 4.24 × 10^{-21}	1.60×10^{-19}	(1.51×10^{-21}) 1.90×10^{-21}	1.15×10^{-20}	1.21×10^{-20}	1.58×10^{-18}	4.24×10^{-20}			1.23×10^{-20}		3.46×10^{-19}	2.65×10^{-20}	1 47 × 10 ⁻¹⁹		1.00×10^{-18}	$9,25 \times 10^{-19}$	
$\sigma_{v_0}(O_2)$ (cm ²)	$ \begin{array}{c} 1.63 \times 10^{-19} \\ 8.15 \times 10^{-20} \\ 3.05 \times 10^{-19} \\ 6.86 \times 10^{-20} \\ 6.57 \times 10^{-20} \end{array} $	1.12×10^{-19} 1.12 × 10^{-19}	1.32×10^{-20}	1.79×10^{-19} 1.32×10^{-20}	4.85×10^{-19}	5.29×10^{-19}	$\begin{array}{c} 5.39 \times 10^{-19} \\ 3.79 \times 10^{-19} \\ 3.60 \times 10^{-19} \\ 2.36 \times 10^{-19} \end{array}$	9.74×10^{-19}	1.17×10^{-18} 1.05×10^{-18}	1.26×10^{-18}	$7.14 \times 10^{-19} \\ 7.14 \times 10^{-19} \\ 7.14 \times 10^{-19} \\ \end{bmatrix}$	6.35×10^{-19} 6.35×10^{-19}	6.35×10^{-19} 7,14 × 10^{-19}	7.14×10 ⁻¹⁹	3.90×10^{-19} 3.90×10^{-19} 3.90×10^{-19}	6.24×10^{-19}	$\begin{array}{c} 6.24 \times 10^{-19} \\ 6.24 \times 10^{-19} \\ 6.94 \times 10^{-19} \end{array}$	6.94×10^{-19}	4.42×10^{-19}	$\{6, 15 \times 10^{-19} \}$ $\{6, 15 \times 10^{-19} \}$	3.58×10^{-19} 5.29 × 10^{-19}	
r_{f}	$\left\{\begin{array}{c} 0.02726\\ 0.01363\\ 0.05110\\ 0.00722\\ 0.00722\end{array}\right.$	0.01180	0.00096	0.01886 0.00096	$\begin{cases} 0.05102 \\ 0.04198 \end{cases}$	0.03883	$ \left\{\begin{array}{c} 0.02821\\ 0.01986\\ 0.01886\\ 0.01236 \end{array}\right. $	0.05102	$\begin{cases} 0.06109 \\ 0.05494 \end{cases}$	0.04392	$\left\{\begin{array}{c} 0.02478\\ 0.02478\\ 0.02478\end{array}\right.$	0.02196	$\left(\begin{array}{c} 0.02196 \\ 0.02478 \\ 0.02478 \end{array} \right)$	0.02478	0.01399	0.02230	$ \left\{\begin{array}{c} 0.02230\\ 0.02230\\ 0.02480 \end{array}\right. $	$\{0.02478$	0.01701	$\left\{ \begin{array}{c} 0.02196 \\ 0.02196 \end{array} \right\}$	0.01278	e 4.01 m length cell.
S Bethke (cm)	5.98×10 ⁻¹⁸ 0.40×10 ⁻¹⁸	9.49×10^{-18}	1.38×10^{-17}	9.49×10^{-18} 1.38 × 10 ⁻¹⁷	9.49×10^{-18}	1.38×10^{-17}	1.91×10^{-17}	1.91×10^{-17}	1.91×10^{-17}	2.87×10^{-17}	2.89×10^{-17}		2.89×10^{-17}		2.80×10^{-17}	2.80×10^{-17}	2.80×10^{-17}	2.80×10^{-17}	2.60×10^{-17}	2.80×10^{-17}	$\begin{array}{c} 2.80 \times 10^{-17} \\ 2.60 \times 10^{-17} \end{array}$	ere obtained with the
$\overline{\Delta v}(cm^{-1})$	33.2	4.9	1.2	15.2	7.9	0.6	32.8	5.1	6.2	0.18	1.97			8.29		0.41	4.69	3 19		0.59	0.12	heses; others w
$\Delta v(cm^{-1})$	28.0 32.6 33.3	46.5 4.9	1.2	15.7 7.5	11.5 3.5	0.6	35.8 17.6 28.3 50.8	5.1	5.4 7.1	0.18	2.01 1.66 2.25	8.45 11.51	11.51 11.34	7.09 7.09	4.49 2.60 4.83	0.41	5.20 4.29 4.59	3.38 3.96	1.70	0.59	0.12	own in parent
$v_{s_i}(cm^{-1})$	54 185.3	54 236.7	54 305.1	54 313.8	54 451.1	55 038.0	55 124.4	55 276.0	55 309.4	56 280.19	56 467.75			90.114.00		56 613.79	56 623.13	56 637 28		56 656.27	56 700.25	length cell are sh
v _{o2} (cm ⁻¹)	54 157.3 54 152.7 54 218.6 54 218.6	54 231.8 54 231.8 54 231.8	54 306.3	54 298.1 54 306.3	54 462.6 54 454.6	55 037.4	55 160.2 55 142.0 55 096.1 55 073.6	55 270.9	55 314.8 55 302.3	55 280.37	56 465.75 56 469.41 56 470.00	56 485.54 56 488.60	56 488.60 56 465.75	56 470.00	56 479.69 56 481.92	56 613.38	56 617.93 56 618.84 56 627.72	56 633.90 56 633.32	56 635.58	56 655.68 56 655.68	56 700 37 56 700.37	ed with the 2.17 m
Line	IR 33R 23R 23R	21P 21P 21P	29P	19P 29P	13P 15R	3P	17P 19R 21R	13P	11P 13R	$7P_{2,3}$	9R1 R2 R3	$_{P_2}^{7P_1}$	9 _{R1}	Ϋ́Ϋ́Ϋ́Ϋ́Ϋ́Ϋ́Ϋ́Ϋ́Ϋ́Ϋ́Ϋ́Ϋ́Ϋ́Υ	اکلار R2	$9P_1$	9P2 P3	${}^{9}R_{2}$	$13P_1$	$_{P_3}^{7P_2}$	$^{3P_{1}}_{11P_{2}}$	/alues obtaine
Band	8-0 9-0	0-6	10-0	0-0 0-0	0-6	10-0	11-0	11-0	11–0	14-0	15-0	15-0			10-01	16-0	16-0	16-0	17–0	16–0	16-0 17-0	Nore: /

lines (1–4). Differences between measured values were not clearly understood before analysis of the rotational structure by Curry and Herzberg (5), Knauss and Ballard (6), and Brix and Herzberg (7), all of whom used continuous light sources and high resolution spectrographs.

Absorption measurements by Watanabe, Inn, and Zelikoff (8) using a hydrogen light source showed an apparent pressure effect due to insufficient resolution. They lead to an order of magnitude of absorption intensity useful for semiquantitative applications. Using a slightly higher resolution and assuming the spectral line width, Ditchburn and Heddle (9) have deduced values of the oscillator strengths of the bands. Such results do not agree with those of Bethke (10), deduced from integrated absorption coefficients measured at high pressure, or with those of Hudson (11), obtained recently at different temperatures. This discrepancy has lead to the conclusion that the line half-widths assumed by Ditchburn and Heddle were too small.

The present article reports measurements of the absorption coefficients of O_2 on more than 30 lines of atomic silicon excited in a microwave discharge tube. Such a light source, already described (12), presents various advantages; the spectral lines are narrow and their wavelengths are known with a high accuracy, the uncertainty being less than 0.0015 Å. In addition, such a source produces practically no continuous radiation.

Experimental

The spectral lines have been isolated in our experiments using a 50 cm focal length vacuum monochromator before entering the absorption cell which was a Pyrex tube closed at each end with a quartz lens. Two cell lengths have been used, 217 and 401 cm. The light intensity was monitored by an E.M.I. 6255 Cd photomultiplier. The output was recorded for various O_2 pressures measured by using mercury manometers and MacLeod gauges.

For the data obtained in the region between the 5–0 and the 17–0 band, Beer's law is obeyed and the values range from 1.5×10^{-22} to 1.6×10^{-18} cm² and are presented in Table I. In a few cases where the wavenumber of an O₂ rotational line differs only by a fraction of a cm⁻¹ from the wavenumber of a silicon line at which the measurement takes place, the absorption crosssection has a negative pressure coefficient. In such cases an inferior limit for the absorption cross-section was obtained by extrapolating to zero pressure. For lower ν' values a positive pressure coefficient appears which has already been reported by others (13, 14). We shall not deal here with this part of our data.

Interpretation

The absorption at 56 700.25 cm⁻¹ of one of the Si lines corresponds to the maximum of the $3P_1$ line of the 16.0 band and the $11P_2$ line of the 17.0 band of O_2 which are blended at 56 700.37 cm⁻¹. We infer, from the integrated absorption coefficients given by Bethke and from the relative rotational population, that these lines have an equivalent rectangular width of 1 cm⁻¹.

This value is not in disagreement with those given by Wilkinson and Mulliken (15), Heddle (16), and Hudson (11). Assuming that this value is valid for the bands that we consider here, we have calculated the maxima of absorption crosssection at the center of the lines which are close to the silicon lines. We have then plotted the ratios of the measured values σ to the sum of the maximum values σ_i versus the distance Δv in cm⁻¹, weighted according to the relation

$$\Delta v \left(\mathrm{cm}^{-1} \right) = \frac{\sum \Delta v_i \sigma_i}{\sum \sigma_i}$$

The graph of Fig. 2 shows the absorption dependence of the distance from the line even at 30 cm^{-1} from the line center. The error bars indicate only the uncertainty on the measured cross-sections. The scattering of the points can be attributed to the uncertainty of the wave-numbers for the bands having ν' values lower than 12, to the assumption of a constant half-width and to the overlapping of the bands which affects Bethke's values.

The solid line represents Rice's formula (17) for the shape of a line broadened by predissociation. To obtain the best fit of this curve with our data the half-width has to be fixed at 1.4 cm^{-1} .

Conclusion

A line half-width of 1.4 cm^{-1} , which must be chosen to obtain the best fit to the data mainly at large values, of Δv where it has the largest influence on the ratio σ/σ_i , indicates that the band oscillator strengths given by Bethke may be too low by as much as a factor of two or three at least for the v' values smaller than 12.

No continuous absorption underlying the bands on which we report in this article can be deduced from our measurements. The data reported here enforce the old idea of Flory (18)



FIG. 2. Ratios σ/σ_i of the measured cross section σ and calculated cross section at the line center σ_i vs. the distance Δv from the line center.

relating to the predissociation of the oxygen molecule in the Schumann-Runge bands.

A synthetic absorption spectrum can be computed using a line half-width of 1.4 cm^{-1} and Bethke's values for the band oscillator strengths. It exhibits minima and maxima of absorption cross-section which lead to unit optical depth in the atmosphere between altitudes at 25 and 120 km. The deduction of precise values will only become possible when the position of the lines and the strength of the bands has been accurately determined.

- 1. H. KREUSLER. Ann. Phys. 6, 412 (1901).
- II. N.REUSLER. AHII. Phys. 0, 412 (1901).
 L. P. GRANATH. Phys. Rev. 34, 1045 (1929).
 H. BUISSON, C. JAUSSERAN, and P. ROUARD. Rev. Optique, 12, 70 (1933).
 A. VASSY. Ann. Phys. Paris, 16, 145 (1941).
 J. CURRY and G. HERZBERG. Ann. Physik, 5 Folge, 19 800 (1934).
- 19, 800 (1934).
- 6. H. P. KNAUSS and S. S. BALLARD. Phys. Rev. 48, 796 (1935).

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- 7. P. BRIX and G. HERZBERG. Can. J. Phys. 32, 110
- P. BRIX and G. HERZBERG. Can. J. Phys. 22, 115 (1954).
 K. WATANABE, E. C. Y. INN, and M. ZELIKOFF. J. Chem. Phys. 21, 1026 (1953).
 R. W. DITCHBURN and D. W. O. HEDDLE. Proc. Roy. Soc. London, Ser. A, 226, 509 (1954).
 G. W. BETHKE. J. Chem. Phys. 31, 669 (1959).
 R. D. HUDSON and V. L. CARTER. Private communication
- munication.
- 12. V. KAUFMAN, L. J. RADZIEMSKI, and K. L. ANDREW. J. Opt. Soc. Am. 56, 911 (1966).
- W. HEILPERN. Helv. Phys. Acta, 14, 329 (1941).
 R. W. DITCHBURN and P. A. YOUNG. J. Atmospheric Terrest. Phys. 24, 127 (1962).
 P. G. WILKINSON and R. S. MULLIKEN. Astrophys. 125 504 (1957).
- P. G. WILKINSON and K. S. MOLLIKEN. ASTOPHYS J. 125, 594 (1957).
 D. W. O. HEDDLE. J. Chem. Phys. 32, 1889 (1960).
 O. K. RICE. Phys. Rev. 35, 1551 (1930).
 P. J. FLORY. J. Chem. Phys. 4, 23 (1936).

Predissociation in N_2 and O_2^1

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An outline discussion is given of some of the complications in the measurement of absorption cross sections which arise from finite instrumental bandwidths. Application is made to data on O2 and N2 from the literature.

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Introduction

In a recent paper (1) we have shown that published absorption cross sections obtained at maxima or minima could be in error by as much as a factor of two, and that published line and band profiles are probably erroneous because of a failure to take into account the effects of finite instrumental bandwidths. In this paper we should like to present some measurements that we have performed on N_2 (2), and O_2 (3) at high resolution, and to show that another consequence of poor resolution is a loss of information regarding the physical processes that accompany the absorption.

Results

Molecular Oxygen

We would like to discuss the molecular oxygen results first. Our initial aim was to study the effect of temperature on the absorption of oxygen in the Schumann-Runge system, at an instrumental bandwidth of 0.075 Å. The results of our study of the dependence of the continuum on temperature were published some time ago (4). The dependence of the band structure is more complicated as is indicated in Fig. 1, which covers the spectral range from 1878 to 1894 Å. All three absorption traces were obtained with the same concentration of oxygen in the absorption cell, thus the effects that one sees are due entirely to the redistribution in the ground vibrational and rotational levels. The curve at 600 °K is of special interest as the fraction of molecules in the v'' = 1 state at that temperature is 0.0234, and yet the 9-1 band obviously gives rise to more absorption than the 6-0 band. This fact reflects the large increase in Franck–Condon factors from the v'' = 0 to the v'' = 1 levels due in part to the relatively large (0.4 Å) difference between the internuclear separations of the ground $X^{3}\Sigma_{g}^{-}$ and the excited $B^{3}\Sigma_{u}^{-}$ electronic states. One of the most important findings of the analysis was that all of the upper vibrational levels from v' = 3 to 17 were subject to predissociation, the individual rotational lines having half-widths varying from 0.5 to 2.3 cm⁻¹. Predissociation in these bands explains the failure, thus far, to observe fluorescence in the Schumann-Runge bands in the upper atmosphere, and provides an additional source of atomic oxygen between 65 and 95 km, both atoms of which will be left in the ³P state with kinetic energies of up to 1 eV. We have recently measured this photodissociation rate in the laboratory, as a function of column height, at both 200 and 300° K and the resulting atomic oxygen production rate for the CIRA 1965 standard atmosphere (12) is shown in Fig. 2. The curve shown with the Schu-

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