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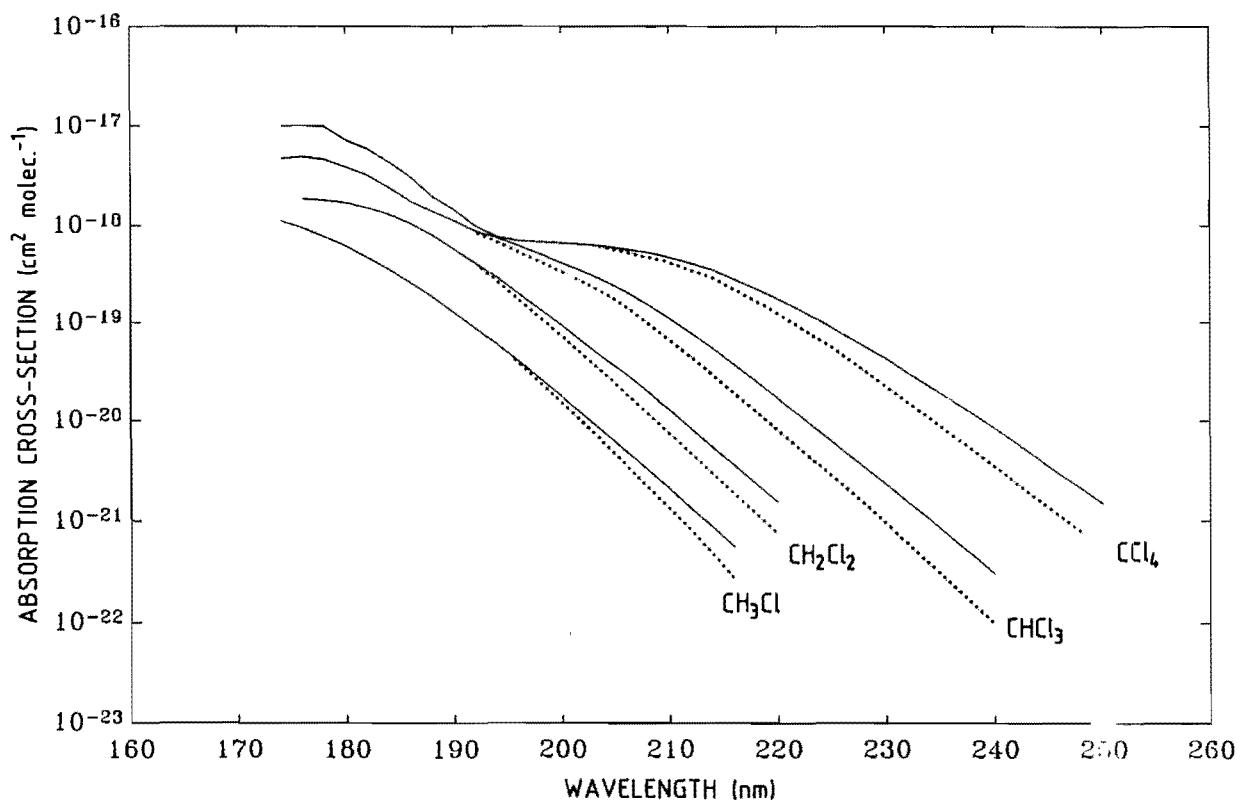


Figure 1 : U.V. Absorption cross-section of chloromethanes
as a function of wavelength

— : T = 295 K
- - - : T = 210 K

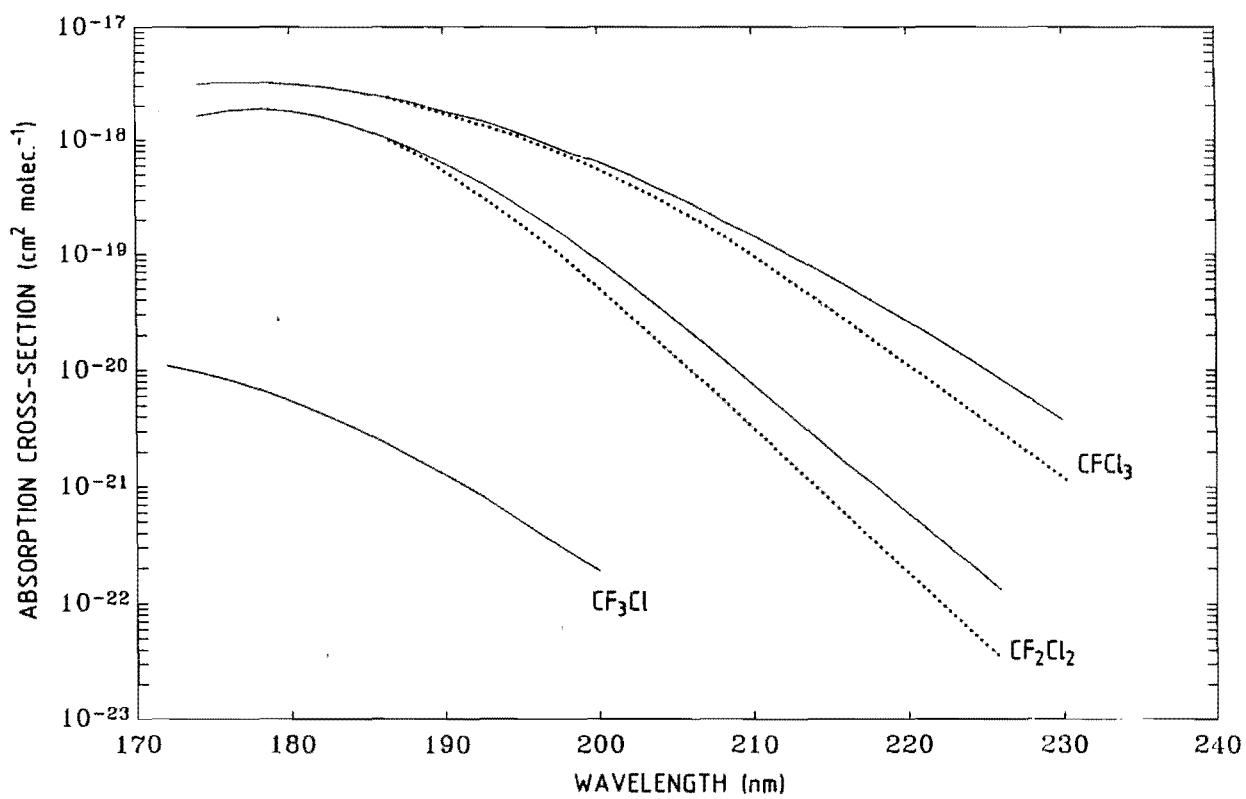


Figure 2 : U.V. Absorption cross-section of chlorofluoromethanes as a function of wavelength

— : T = 295 K
- - - : T = 210 K

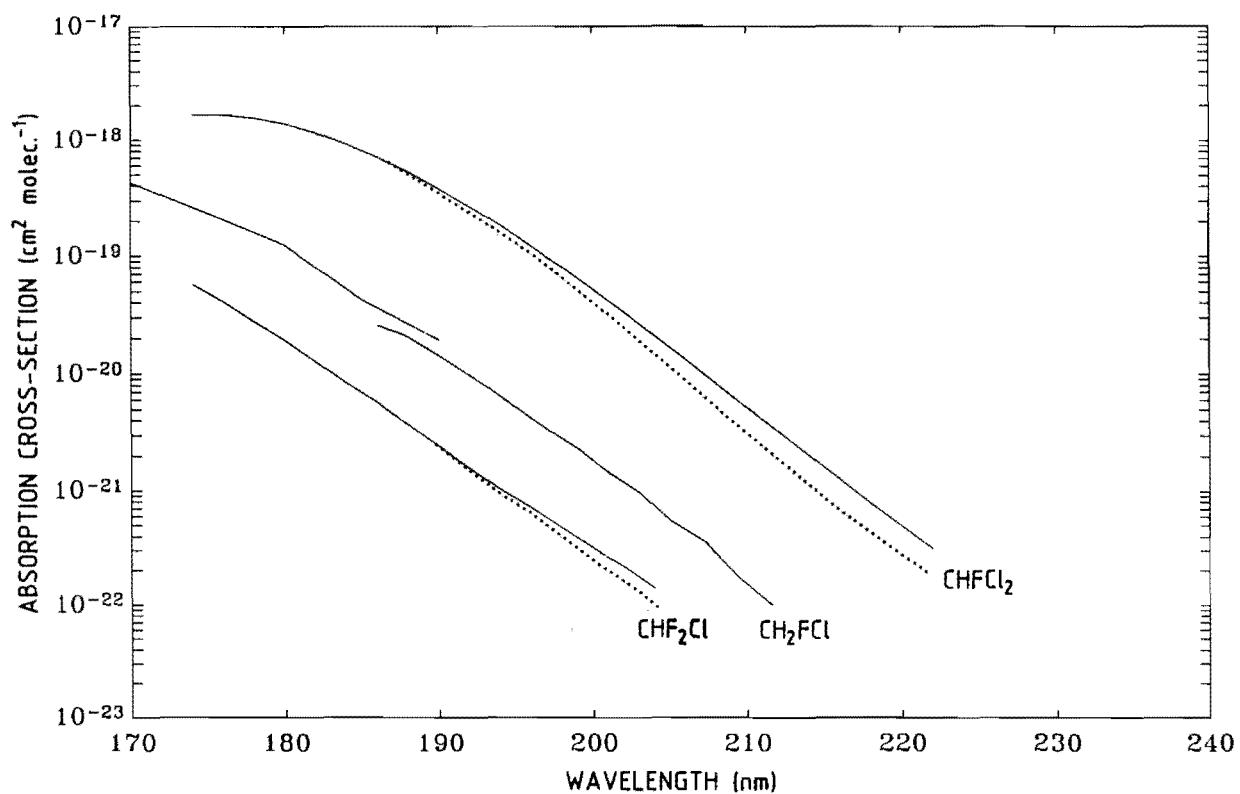


Figure 3 : U.V. Absorption cross-section of hydrogenochlorofluoromethanes as a function of wavelength

— : T = 295 K
 - - - : T = 210 K

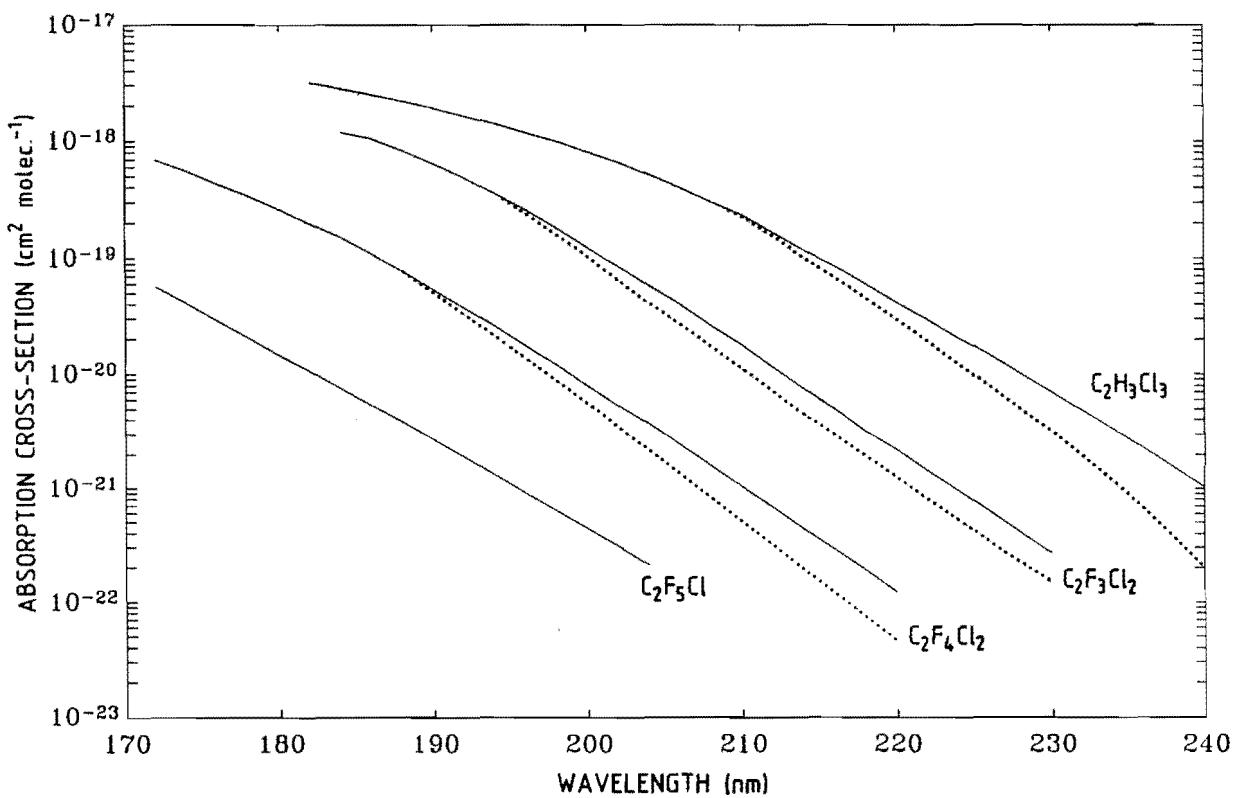


Figure 4 : U.V. Absorption cross-section of chloro and chlorofluoro ethanes as a function of wavelength.

— : T = 295 K
- - - : T = 210 K

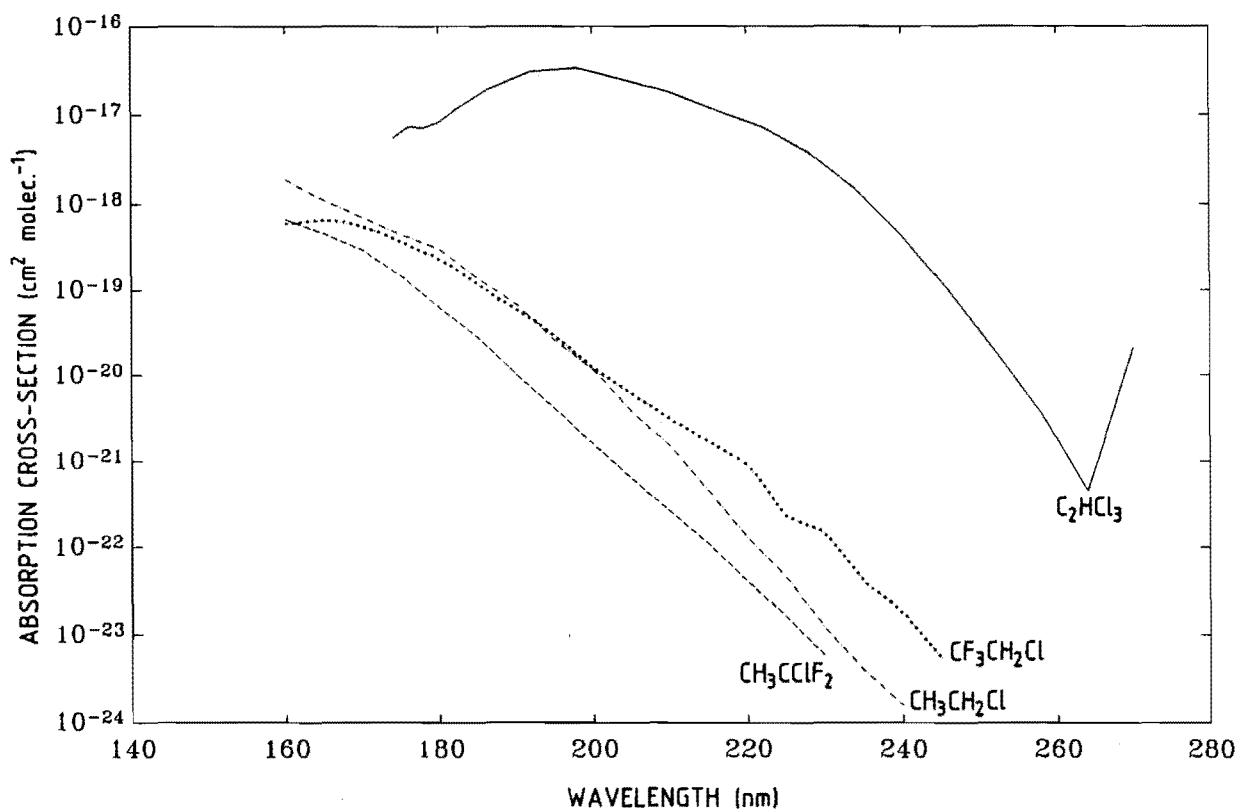


Figure 5 : U.V. Absorption cross-section of chlorofluoroethanes and ethylenes at ambient temperature as a function of wavelength.

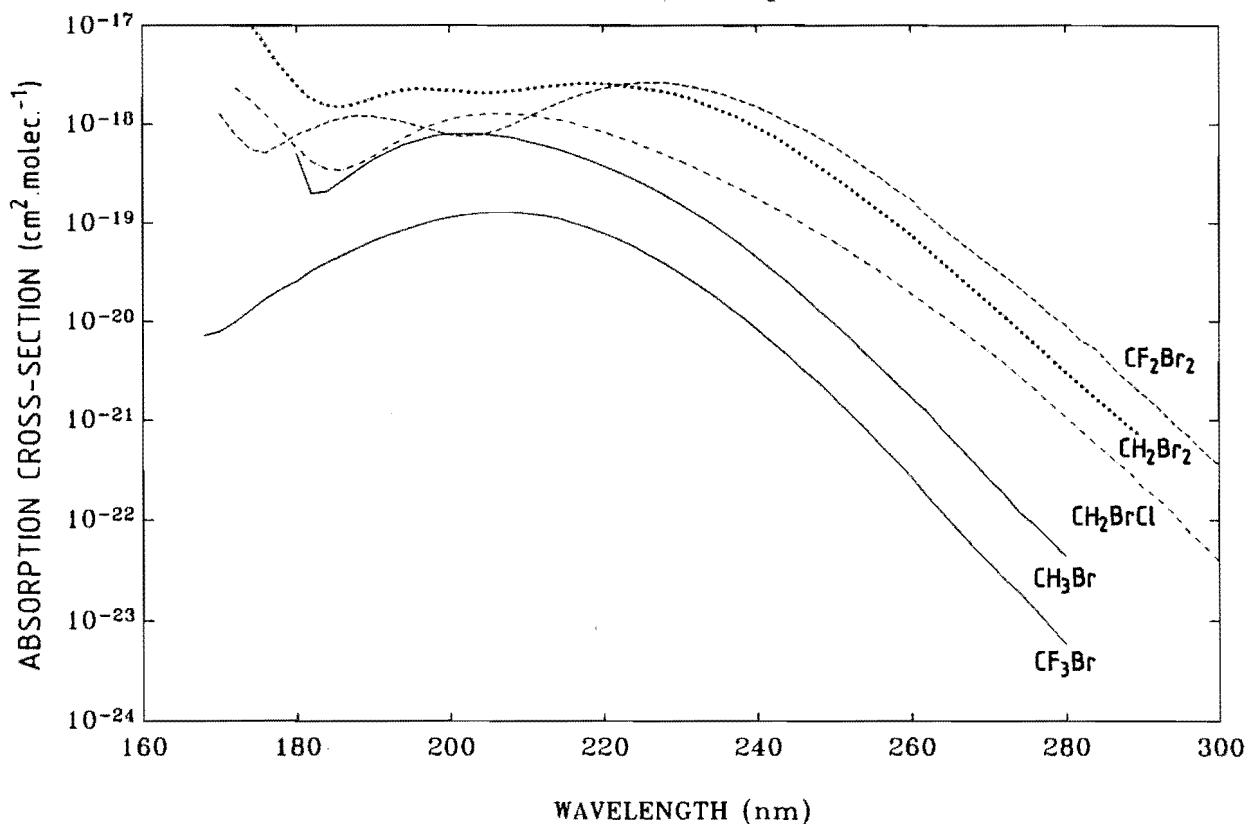


Figure 6 : U.V. Absorption cross section of bromomethanes at ambient temperature as a function of wavelength.

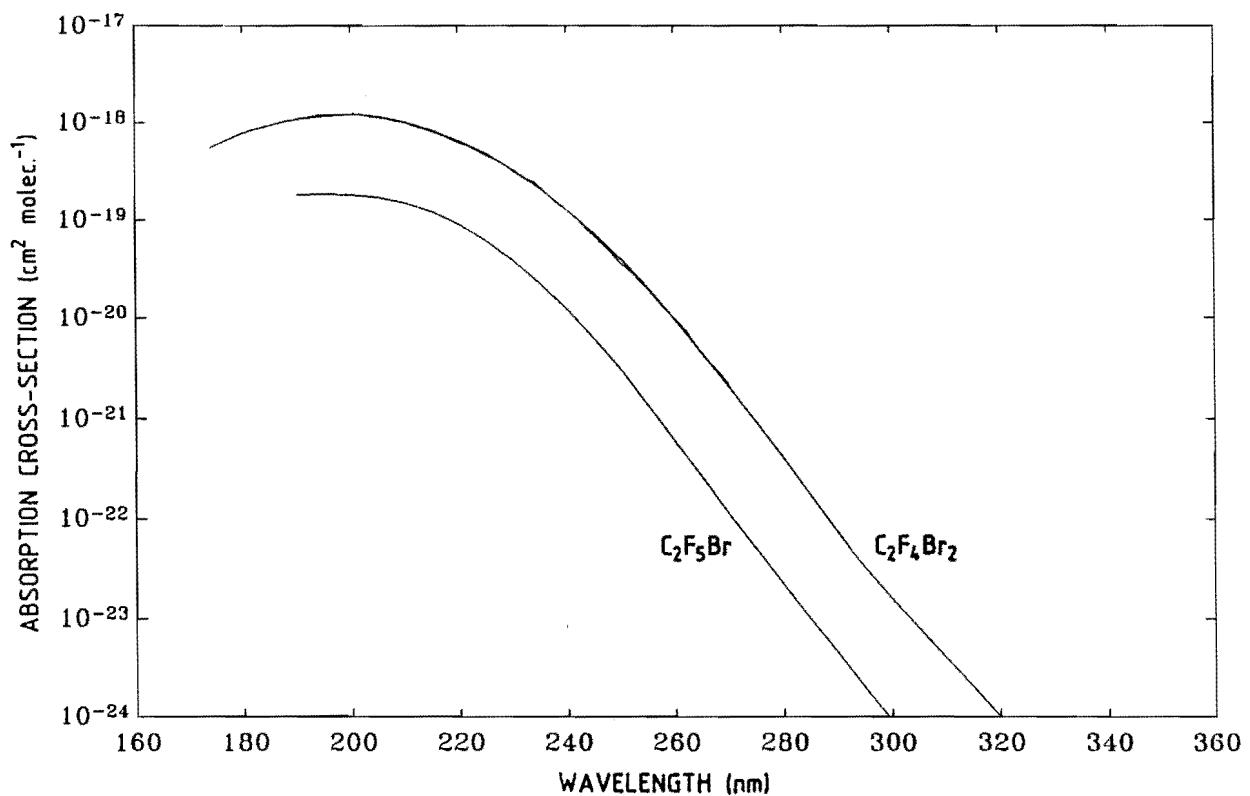


Figure 7 : U.V. Absorption cross-section of bromoethanes at ambient temperature as a function of wavelength.