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Review

Laboratory measurements and calculations of line shape parameters of the H₂O–CO₂ collision systemL. Régalia^{a,*}, E. Cousin^a, R.R. Gamache^b, B. Vispoel^b, S. Robert^c, X. Thomas^a^aGroupe de Spectrométrie Moléculaire et Atmosphérique, UMR CNRS 7331, UFR Sciences Exactes et Naturelles, Moulin de la Housse, BP 1039 – 51687 Reims Cedex 2, France^bDepartment of Environmental, Earth, and Atmospheric Sciences, University of Massachusetts Lowell, 1 University Avenue, Lowell, MA 01854, USA^cPlanetary Aeronomy, Royal Belgian Institute for Space Aeronomy, 3 Avenue Circulaire, 1180 Brussels, Belgium

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ABSTRACT

For decades, the remote sensing measurements have been made in planetary atmospheres in the Solar System and beyond. As the performance of the space instruments improves, the atmospheric science community is more and more in need of accurate spectroscopic data. The current databases offer some parameters for non-Earth atmospheres but are far from complete for all situations. For example, measured H₂O line parameters in CO₂-rich atmospheres such as Mars and Venus are missing while they are of prime importance to learn about the evolution of the atmospheres. New Fourier Transform Spectrometer spectra were recorded respectively around 2.7 and 6 μm, using a Connes' type FT spectrometer built in Reims. The spectra were analysed using a multispectrum fitting procedure to obtain the line-shape parameters of H₂O broadened by CO₂. Modified Complex Robert-Bonamy calculations of the half-width, line shift, and their temperature dependence were made in the spectral region from 1300 to 5000 cm⁻¹. The measurements and calculations are presented and compared with data available in the literature.

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1. Introduction

Water is a key molecule in the search for life on other planets. The origins of water on Earth are not yet well understood. In order to gain insights into this topic, the Solar system must be considered as a whole and the water contents of the different planetary and cometary bodies characterized. This study includes the understanding of the water cycle, the determination of the volume mixing ratio and the distribution with altitude of the different isotopologues of water. In particular, the ratio of deuterium to hydrogen (D/H) is a key diagnostic to characterizing the Solar system's evolution.

Several space instruments have flown or are flying around Mars and Venus. While these planets are our closest neighbors, they have a very different atmospheric structure and composition. Nadir viewing, provided that the spectral resolution and the SNR technique are adequate, and solar occultation viewing allow probing an atmosphere layer by layer to give vertical profile information. The solar occultation technique has the advantage of extremely localized sensitivity to the true atmospheric state (i.e. measured by Averaging Kernels), highly peaked on the impact tangent height.

These profiles allow isotopic ratios, such as D/H, to be studied. Considering Mars, the ratio is an important evolutionary tracer for the atmosphere, surface, and their interactions in the climate system. The isotopic ratios are very useful because they are not as affected by modification related to the chemistry occurring in Mars' atmosphere as elemental abundances are [1]. Investigations of the D/H ratio with altitude has led to interesting results [2–7]. Studying the spatial distribution of the D/H ratios can also generate interesting data. Villanueva et al. [8] used these 2D maps to study strong water isotopic anomalies in the Martian atmosphere; Good et al. [9] looked at D/H isotope ratios in the global hydrologic cycle; and Yang et al. [10] modeled the D/H ratio of water in the solar nebula during its formation and evolution. These results could be refined if better line parameters were used, especially H₂O broadened by CO₂ line shape parameters. The current parameters used by the planetary community rely on empirical adaptation of other line lists or on scaling other broadening species data. In this study, we use our laboratory set-up and multispectrum procedure to provide measured line shape parameters and make state-of-the-art Modified Complex Robert-Bonamy (MCRB) calculations of the half-width, line shift, and their temperature dependence for the H₂O–CO₂ collision system.

Two spectral ranges of interest were identified around 2.7 and 6 μm and spectra recorded. The 6 μm band was analyzed previously by Brown et al. [11] in 2007, while the 2.7 μm band was the

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Table 1

Experimental conditions of recorded spectra with the Connes' type FTS ("DDM max" is the maximum of path difference for the interferogram measurement).

Spectral region 2.7 μm		iris radius (aperture) = 2.25 mm			
Cell length (cm)	DDM max (cm)	Total Pressure in Torr	P(H ₂ O) in Torr	P(CO ₂) in Torr	Temperature in Kelvin
13.6	46	504.2	2.05	502.2	295.5
	61	300.9	1.22	299.7	295.5
	88	99.3	0.40	98.9	295.6
417.2	35	503.5	2.03	501.5	292.7
	41	300.5	1.21	299.3	292.9
	59	99.9	0.40	99.5	293.2
817.8	29	507.1	2.10	505.0	293.3
	33	300.7	1.24	299.4	293.4
	59	96.6	0.40	96.2	293.3
Spectral region 6 μm		iris radius (aperture)= 2.5 mm			
Cell length (cm)	DDM max (cm)	Total Pressure in Torr	P(H ₂ O) in Torr	P(CO ₂) in Torr	Temperature in Kelvin
13.6	46	502.9	2.07	500.8	295.4
	69	298.3	1.23	297.1	295.4
	70	100.3	0.41	99.9	295.5

target of a similar study, here in Reims, some twenty years ago [12]. The spectrometer in Reims now has a better signal to noise ratio (~ 1000) than previously [12] and the software developed to analyse the spectra in a multispectrum procedure [13] allows the retrieval of more consistent line shape parameters. For the deuterated species of water, HDO, measurements were made on transitions in the ν_2 , ν_1 , and ν_3 bands [14–16] and MCRB calculations were made for the rotation, ν_2 , ν_1 , and ν_3 bands [14–17]. In the present study, the newly measured spectra were analyzed in order to determine accurate line shape parameters of H₂O broadened by CO₂. We compared our measured results to the data available in the literature and with the MCRB calculations described below.

Section 1 describes the experimental conditions of the measured spectra and explains the line parameters retrieval procedure. Section 2 presents the theory and the calculations made. Finally, we will present our line shape measurements and calculations and compare them with literature values.

2. Experimental conditions

2.1. Spectra recording

Sixteen spectra were measured at room temperature with the Connes' type Fourier Transform Spectrometer (FTS) [18–20] built in the Group of Molecular and Atmospheric Spectrometry laboratory (GSMA) in Reims, France. This instrument had a 3-meter maximum path difference allowing a non-apodized resolution of 0.0017 cm^{-1} . Table 1 summarizes the recording conditions of the spectra in the two studied spectral regions. In the 2.7 μm region, all spectra were recorded with the following optical setup: a CaF₂ beam splitter, some lenses and BaF₂ windows and two InSb detectors, while in the 6 μm region we used two HgCdTe detectors. We always checked in spectral regions near 10 μm if the sample or the window emission could perturb the study but in the stated situation, we didn't observe anything. In addition to this, in the 6 μm spectral region to eliminate some waves on the 100% of transmission of the spectra due to optical windows in the optical path, we had to divide by a spectrum recorded with an empty cell. We used two different optical spectral filters to select the recording spectral regions, the transmission windows are $1350 - 2300 \text{ cm}^{-1}$ and $3060 - 4370 \text{ cm}^{-1}$ respectively for the 6 μm and 2.7 μm spectral region. During the experiment, the absorption path was maintained under vacuum and the pressure and the temperature were continuously monitored. Two absorption cells were used: a single-path cell with an optical path length of 13 cm and a 1-meter White cell

[21] to achieve absorption path lengths of 4 and 8 m. As shown in Table 1, for all spectra, the partial pressure of water vapor was kept very low in order to neglect as much as possible the influence of the self-broadening due to the H₂O pressure in the determination of the line-shape parameters. In addition to the spectra listed in Table 1, we recorded a spectrum for each length of the cell, with only a low H₂O pressure (1 Torr). These spectra allowed us to check the instrumental line shape and were used to calibrate the wavenumber scale of the spectra described in Table 1. For the spectral calibration, the line positions given in Loos et al. [22] were used as reference. This calibration is not crucial, as the multispectrum procedure allows discrimination of the different shifts for a given recording cell. Indeed, as the parameters are fitted simultaneously on several spectra in different experimental conditions, the shift due to the optical set-up is a parameter with the same value for all spectra, while the shift due to the pressure of the gas (absorbing or foreign) is depending on the studied spectrum.

The carbon dioxide buffer gas was supplied by Air Liquide company with a stated purity of 99.9999%. The pressure was measured with an uncertainty less than 0.3% using two MKS Baratron manometers with 10 and 1000 Torr full scale. The temperature was measured with a platinum probe with an uncertainty less than 0.5 K.

The Fig. 1 shows an example of the record spectra in the 2.7 μm spectral region

2.2. Line parameters retrieval

We used the multispectrum procedure developed at GSMA [13] to determine the line parameters. First, in order to minimize the effect of the apparatus function on the retrieved parameters, the experimental value of the Michelson's interferometer aperture was determined using spectra recorded at low pressures (not reported in the Table 1). The iris radius values of the FTS are indicated in Table 1. We did this verification even if the choice of the radius value is less critical than in the Doppler regime, since the H₂O spectra were broadened by a pressure of CO₂ from 100 Torr to 500 Torr. The verification was done on several lines taken in the whole spectral region allowing a check of the nominal iris radius that can be used. No specific signature appeared on the residuals of the fit during this verification. The nominal iris radius value was then used in the following analysis.

The multispectrum procedure enabled us to obtain simultaneously the position of the line, the line strength, the half-width and the line shift. An example of the multispectral fitting procedure

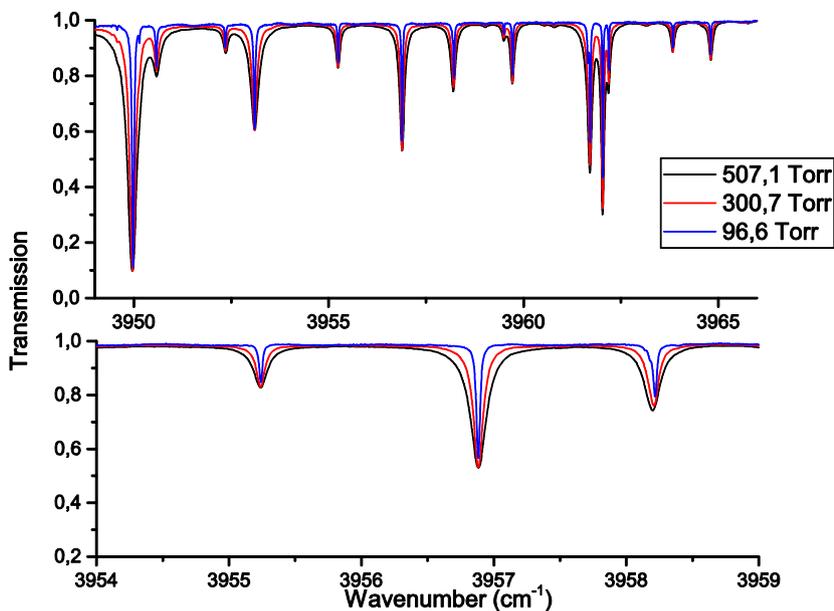


Fig. 1. Example of recorded spectra in the 2.7 μm spectral region around 3956 cm^{-1} , with an absorption path length of 817.8 cm (see Table 1).

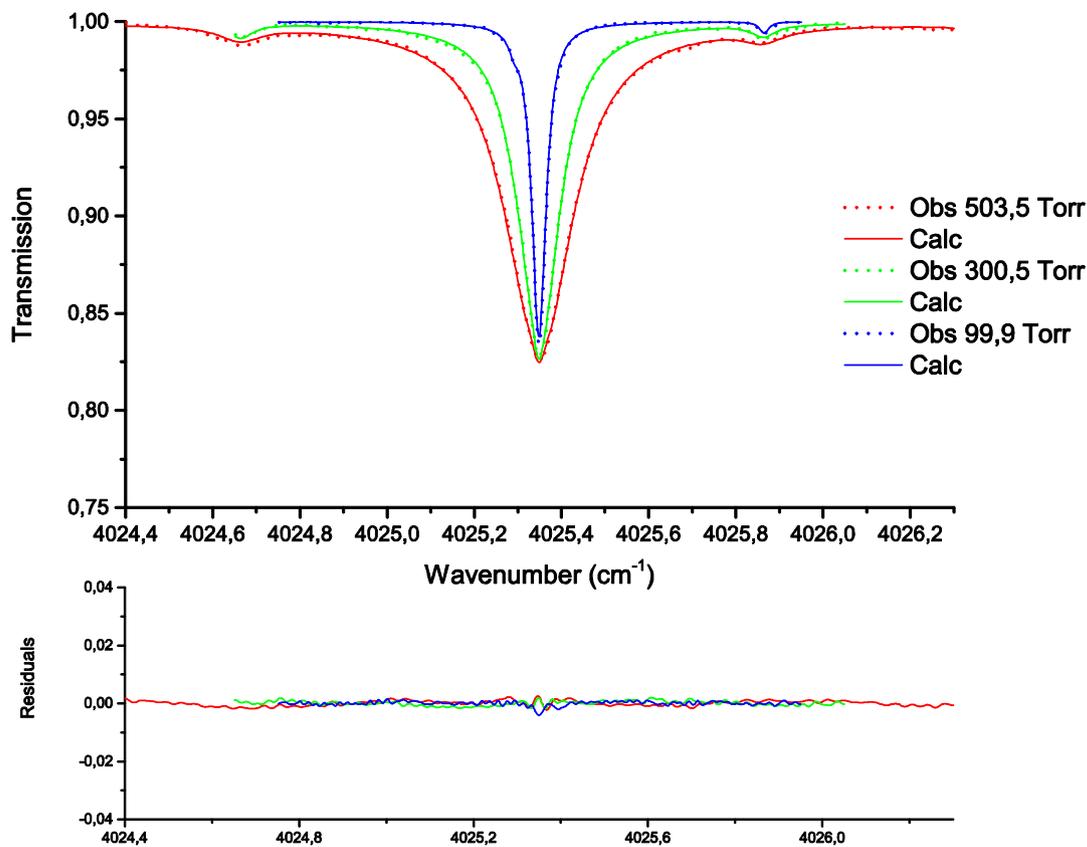


Fig. 2. Example of the fit with the multispectrum procedure of a H_2O transition broadened by CO_2 in the 2.7 μm spectral region. The three spectra have an absorption path length of 417.2 cm (see Table 1).

is shown in Fig. 2. The figure shows the simultaneous fit of three spectra with an absorption path length of 417.2 cm and different pressures (see Table 1) in the upper panel and the residuals of the fit in the lower panel. Just over ten isolated lines were fit using two different values of self-broadening coefficients (broadening due to the pressure of H_2O) in order to assess the impact of this parameter. A modification of 20% of the fixed value of self-broadening coefficient does not change the retrieved CO_2 -

broadening parameter. Therefore, the self-broadening coefficients were fixed to the values of the HITRAN2016 database edition [23]. The impact of the self-shift parameter was also negligible.

3. Theory

For the collision system under consideration, $\text{H}_2\text{O}-\text{CO}_2$, and temperatures of the study, quantum dynamical methods are too

slow to be practical even if adequate potential energy surfaces were available. Thus, the calculations of the half-width and line shift for the H₂O–CO₂ collision system were done using the semi-classical Modified Complex Robert-Bonamy formalism [24,25] (MCRB). The half-width and the line shift are given by the real and imaginary parts, respectively, of the expression

$$(\gamma + i\delta)_{f \leftarrow i} = \frac{n_2}{2\pi c} \int_0^\infty v f(v) dv \int_0^\infty 2\pi b \times \left[1 - e^{-i\langle S_1 + \text{Im}(S_2) \rangle_{J_2}} e^{-\langle \text{Re}(S_2) \rangle_{J_2}} \right] db. \quad (1)$$

In this expression, n_2 is the number density of bath molecules, c is the speed of light, v is the relative velocity, $f(v)$ is the Maxwell-Boltzmann distribution of velocity, $\langle \dots \rangle_{J_2}$ is an average over the states of the bath molecules, and b is the impact parameter. S_1 (imaginary) and S_2 (complex) are the first and second order terms in the successive expansion of the Liouville scattering matrix, which depend on the ro-vibrational states in the optical transition, the associated collisionally induced jumps from these states, on the intermolecular potential, and the characteristics of the collision dynamics. Note, semi-classical refers to the fact that the internal structure of the colliding molecules is treated quantum mechanically and that the dynamics of the collision process are treated by classical mechanics. Thus, in an optical transition from $f \leftarrow i$, the active molecule will undergo collisions with the bath molecules, for which the trajectories are determined by classical mechanics. In these collisions, the initial and final states of the radiating molecule, i and f , will undergo collisionally induced transitions to states i' and f' , interrupting the radiation and causing collisional broadening and shifting. The states i' and f' are called *collisionally connected states* and are given by selection rules determined by the wavefunctions and intermolecular potential. Given the large number of terms in the intermolecular potential, there are many states, i' or f' , that are possible. The quantum mechanical components of the calculation are the energy of the states involved in the collisionally induced transitions and the wavefunctions for the states, which are used to determine reduced matrix elements (RME), which give the probability of a collisionally induced transition. A recent study [26] showed that MCRB calculations should use the ground state RMEs for all vibrational states.

The intermolecular potential used in the calculations was comprised of electrostatic terms, the induction and London dispersion terms, and an atom-atom component expanded to 20th order and rank 4. The atom-atom parameters were adjusted to bring the MCRB calculations in agreement with measurements of Tretyakov and Koshelev [27]. The final atom-atom parameters (called pot 23) are a change of $\epsilon_{\text{H-O}}$, $\epsilon_{\text{H-C}}$, $\epsilon_{\text{O-O}}$, $\epsilon_{\text{O-C}}$ by -30% , a change of $\sigma_{\text{H-O}}$ by -22.5% and a change of $\sigma_{\text{H-C}}$, $\sigma_{\text{O-O}}$, $\sigma_{\text{O-C}}$ by -25% from the combination rule [28] values. All other parameters are the recommended values from the literature: The dipole moment of water and its vibrational dependence are taken from Shostak and Muentner [29]. The quadrupole moments come from Flygare and Benson [30]. The polarizability and its vibrational dependence are given by Luo et al. [31]. The ionization potential of water is taken from Ref. [32] and is a vibrationally-independent value. For CO₂, the quadrupole moment is from the work of Graham et al. [33], the value of the polarizability is from Bogaard and Orr [34], and the ionization potential is from Tanaka et al. [35]. The numerical values are given in Refs. [36] and [11], for H₂O and CO₂, respectively.

The MCRB calculations included all real and imaginary components, atom-atom component expanded to high order and rank, trajectories determined solving Hamilton's equations, and explicit velocity averaging over the Boltzmann distribution.

Table 2

The number of lines in each cold band from 1300 to 5000 cm⁻¹ for the three principal isotopologues of H₂O.

Band	H ₂ ¹⁶ O	H ₂ ¹⁸ O	H ₂ ¹⁷ O
0 0 0 ← 0 0 0	112	7	0
0 1 0 ← 0 0 0	2326	1297	915
0 2 0 ← 0 0 0	2150	971	493
1 0 0 ← 0 0 0	2479	1320	781
0 0 1 ← 0 0 0	2950	1593	1096
0 1 1 ← 0 0 0	424	165	94
0 3 0 ← 0 0 0	783	368	86
1 1 0 ← 0 0 0	383	163	57
total	11,607	5884	3522

4. MCRB calculations

Using HITRAN2016, all cold-band transitions for H₂¹⁶O, H₂¹⁸O, and H₂¹⁷O in the range 1300–5000 cm⁻¹ with $J'' \leq 22$ and with full quantum assignments (note, in the UCL *ab initio* data [37] on HITRAN only J and symmetry are good quantum numbers, so some transitions do not have all quantum numbers.) were extracted. The number of lines in each cold band is listed in Table 2. The MCRB formalism includes the vibrational dependence of the line shape parameters, thus calculations were made for each band listed in Table 2. The calculations of the half-width and line shift were made at 13 temperatures from 200 to 3000 K in order to determine the temperature dependence of these parameters. For each band, the unique rotational transitions were determined for the three principal isotopologues; then calculations were made for the H₂¹⁶O isotopologue. These half-width and line shift values were used for the three isotopologues of this study since the line shape parameters are roughly equal for all three. The last step was to use the line shape parameter data at the 13 temperatures to determine the temperature dependence by the Gamache-Vispoel model [38]. These data were used to create an improved line file for applications for water vapor in CO₂-rich atmospheres. The previous H₂O–CO₂ line file [17] neglects vibrational dependence for H₂¹⁶O, H₂¹⁸O, and H₂¹⁷O transitions and uses the rotation band half-widths for all transitions. The line shifts, due to their strong vibrational dependence, were from a semi-empirical algorithm [39] or from scaling.

5. Results and comparisons

5.1. Spectral region around 6 μm

In this spectral region, line half-widths and shifts of H₂O broadened by CO₂ were compared to previous laboratory measurements of Brown et al. [11] for the ν_2 fundamental band and to the MCRB calculations done here. The data sets are shown in Figs. 3 and 4 respectively for the half-widths and the pressure-induced line shift. In total 105 lines were analyzed. Shown in Fig. 3 are the measured half-widths with error bars (blue solid triangle symbols), the MCRB calculated half-widths shifted slightly to the left (black open diamond symbols), and the measured half-widths from Brown et al. [11] with error bars shifted slightly to the right (red open square symbols) versus wavenumber of the transition. Comparing the current measurements and calculations for 105 transitions gives an average percent difference (APD) of -2.9% and a standard deviation (SD) of 4.9% . Comparing our measurements with those of Brown et al. for 100 transitions gives an APD of -3.2% and a SD of 3.0% . Comparing the measurements of Brown et al. with our MCRB calculations for 100 transitions gives an APD of 0.17% and a SD of 5.5% . Thus, the half-width data of these two investigations and the calculations are in good agreement.

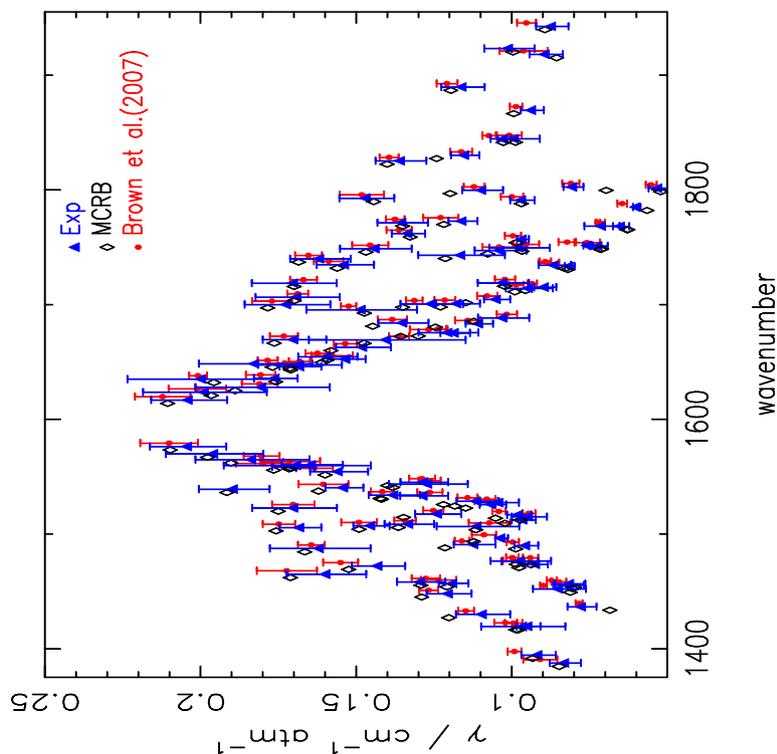


Fig. 3. CO₂-broadened half-widths ($\text{cm}^{-1} \text{atm}^{-1}$) of H₂¹⁶O transitions in the 6 μm region versus wavenumber of the transition. Shown are measurements (Exp) done in this work (blue solid triangle symbols), MCRB calculations from this work (black open diamond symbols), and the measurements of Brown et al. [11] (red solid circle symbols). The error bars given for the measurements correspond to the 2σ statistical uncertainty provided by the fit. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

In Fig. 4, the measured line shifts with error bars (blue solid triangle symbols), the MCRB calculated line shifts shifted slightly to the left (black open diamond symbols), and the measured line shifts from Brown et al. [11] with error bars shifted slightly to the right (red open square symbols) are plotted versus wavenumber of the transition. A comparison of our measurements and calculations for 105 transitions shows an average deviation of $-0.0043 \text{ cm}^{-1} \text{atm}^{-1}$ and a standard deviation of $0.0048 \text{ cm}^{-1} \text{atm}^{-1}$. Comparing our shift measurements with those of Brown et al. for 100 transitions gives an average deviation of $-0.0035 \text{ cm}^{-1} \text{atm}^{-1}$ and a standard deviation of $0.0029 \text{ cm}^{-1} \text{atm}^{-1}$. Comparing the measurements of Brown et al. with our MCRB shift calculations for 100 transitions gives an average deviation of $-0.00065 \text{ cm}^{-1} \text{atm}^{-1}$ and a standard deviation of $0.0040 \text{ cm}^{-1} \text{atm}^{-1}$. Again, showing excellent agreement between this study and that of Brown et al.

Brown et al. [8] presented Complex Robert-Bonamy calculations, which demonstrate that, when vibrational dependence is ignored, pairs of transitions with the same rotational quantum numbers interchanged have exactly the same half-width values and the line shifts have exactly the same magnitude with opposite signs. For asymmetric rotors, Brown et al. extended this property for pairs to an analysis on the parameters K_m and J_m , defined as the maximum values of K_a and J , respectively, in the transition quanta. Brown et al. also define τ' and τ'' as $\tau = K_a + K_c - J$ for use in their analysis. We observed similar behaviors as that seen in Table 8 of Brown et al. [11]. Examples are given in Table 3, where pair transitions are shown for the measured and calculated half-widths and line shifts. Table 3 shows that as a function of J_m and K_m the half-widths are nearly equal and the line shifts approximately equal but opposite in sign. A good example is the sequence for $J_m = 4$ and $K_m = 2$. The exception is when the line shift is very

small, for which the measurements and calculations are difficult to make.

The good agreement with the previous measurements of Brown et al. [11] validates the choice of the experimental conditions for the measured spectra, as well as the intermolecular potential used in the MCRB calculations.

5.2. Spectral region around 2.7 μm

As indicated in Table 1, for this spectral region, we recorded nine spectra with 2 different cells. We fitted simultaneously the spectra with the same cell length, it allows us to obtain lines parameters for transitions in a line intensity range from 3.2×10^{-22} to $9.6 \times 10^{-20} \text{ cm}^2/\text{mole}$: 60 transitions belong to the ν_3 band, 23 to the $2\nu_2$ band and 15 to the ν_1 band. We did not find recent laboratory measurements for this spectral region in the literature. We compared our measurement and MCRB calculation results to the data from the Mars database [17]. Fig. 5 shows the half-widths in $\text{cm}^{-1} \text{atm}^{-1}$ from our measurements with error bars (blue solid triangle symbols), our MCRB calculations (black open diamond symbols), and the previous calculations of Gamache et al. [17] (red open square symbols) versus a line count. The data are arranged such that the first 16 points are ν_1 band data, points 17–40 are $2\nu_2$ band data, and points 41–105 are ν_3 band data. The statistics for the agreement are given in top part of Table 4. The MCRB calculations made here include the vibrational dependence that was neglected in Ref. [17] and the intermolecular potential was optimized for the H₂O–CO₂ collision system as discussed above. The result is that the calculations made here agree well with measurements with an APD of -1.1 , -1.8 , and -3.1 for the ν_1 , $2\nu_2$, and ν_3 bands, respectively. The calculations that neglect the vibrational dependence have -13.8 , -12.0 , and -11.8 APDs for the same data.

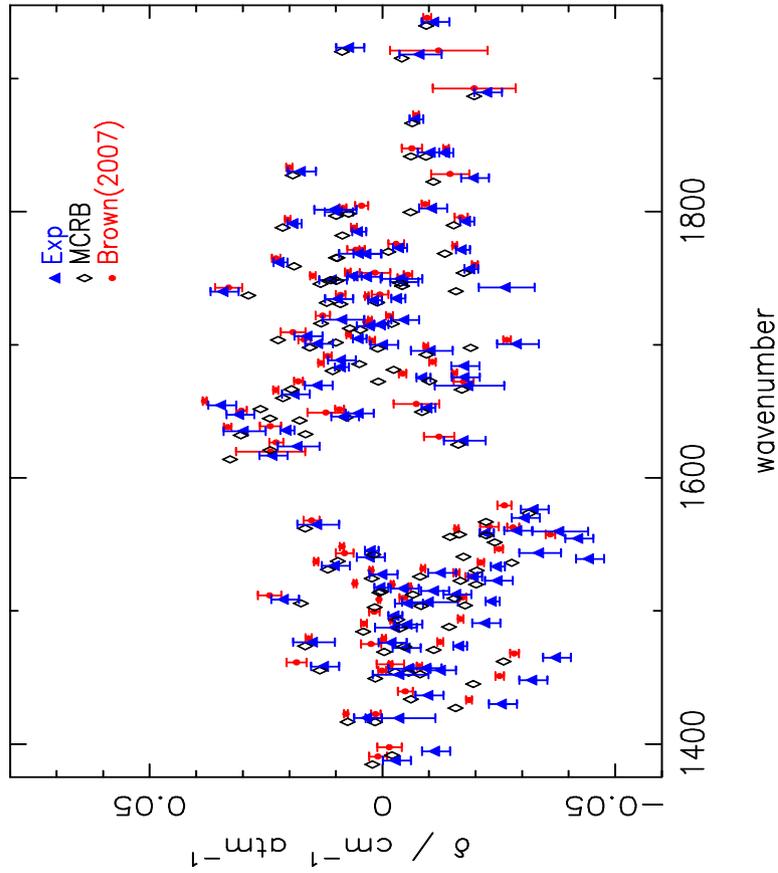


Fig. 4. CO₂ pressure-induced line shifts (cm⁻¹ atm⁻¹) of H₂¹⁶O transitions in the 6 μm region versus wavenumber of the transition. Shown are measurements (Exp) from this work (blue solid triangle symbols), MCRB calculations from this work (black open diamond symbols), and the measurements of Brown et al. [11] (red solid circle symbols). The error bars given for the measurements correspond to the 2σ statistical uncertainty provided by the fit. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

Table 3

Example of measured CO₂ broadened H₂O coefficients for “pair” transitions in the spectral region around 6 μm.

Rotational assignment	Jm	Km	τ'	τ"	γ _{Exp.} cm ⁻¹ /atm	γ _{MCRB} cm ⁻¹ /atm	δ _{Exp.} cm ⁻¹ /atm	δ _{MCRB} cm ⁻¹ /atm
1 1 0 ← 1 0 1	1	1	0	0	0.2037	0.2106	0.02340	0.03271
1 0 1 ← 1 1 0	1	1	0	0	0.2040	0.2098	-0.03270	-0.03144
4 4 0 ← 3 3 1	4	4	0	1	0.0989	0.0988	-0.01060	-0.00610
3 3 1 ← 4 4 0	4	4	1	0	0.0963	0.0980	-0.00393	0.00157
4 4 1 ← 3 3 0	4	4	1	0	0.1025	0.1027	-0.01370	-0.00934
3 3 0 ← 4 4 1	4	4	0	1	0.0946	0.0987	0.00312	0.00747
4 3 1 ← 3 2 2	4	3	0	1	0.1094	0.1198	0.00948	0.00990
3 2 2 ← 4 3 1	4	3	1	0	0.1092	0.1203	-0.02590	-0.01575
4 2 2 ← 4 1 3	4	2	0	0	0.1681	0.1712	0.03050	0.02413
4 1 3 ← 4 2 2	4	2	0	0	0.1689	0.1715	-0.03820	-0.02234
3 1 2 ← 4 2 3	4	2	0	0	0.1428	0.1524	-0.00528	-0.00038
4 2 3 ← 3 1 2	4	2	0	0	0.1535	0.1561	-0.00341	0.00110
4 2 3 ← 4 1 4	4	2	0	0	0.1104	0.1245	0.00870	0.01069
4 1 4 ← 4 2 3	4	2	0	0		0.1265		-0.01342
3 2 1 ← 2 1 2	3	2	0	1	0.1615	0.1685	0.03390	0.02876
2 1 2 ← 3 2 1	3	2	1	0	0.1595	0.1711	-0.03750	-0.02604
4 0 4 ← 3 1 3	4	1	0	1	0.1182	0.1357	-0.00883	0.00092
3 1 3 ← 4 0 4	4	1	1	0	0.1236	0.1348	0.00028	0.00026
3 2 1 ← 3 1 2	3	2	0	0	0.1679	0.1710	0.00803	0.01775
3 1 2 ← 3 2 1	3	2	0	0	0.1641	0.1708	-0.02920	-0.01647
2 0 2 ← 1 1 1	2	1	0	1	0.1801	0.1889	-0.01770	-0.01626
1 1 1 ← 2 0 2	2	1	1	0	0.1833	0.1902	0.01380	0.01658
6 2 5 ← 5 1 4	6	2	1	0	0.1332	0.1326	0.02190	0.01900
5 1 4 ← 6 2 5	6	2	0	1	0.1201	0.1289	-0.03240	-0.01947
7 1 7 ← 6 0 6	7	1	1	0	0.0862	0.0829	0.00939	0.01197
6 0 6 ← 7 1 7	7	1	0	1	0.0818	0.0805	-0.01280	-0.00807
7 0 7 ← 6 1 6	7	1	0	1	0.0814	0.0820	0.00160	0.00900
6 1 6 ← 7 0 7	7	1	1	0	0.0812	0.0799	-0.00583	-0.00562
8 1 8 ← 7 0 7	8	1	1	0	0.0755	0.0712	0.00601	0.01111
7 0 7 ← 8 1 8	8	1	0	1	0.0774	0.0685	-0.01010	-0.00614

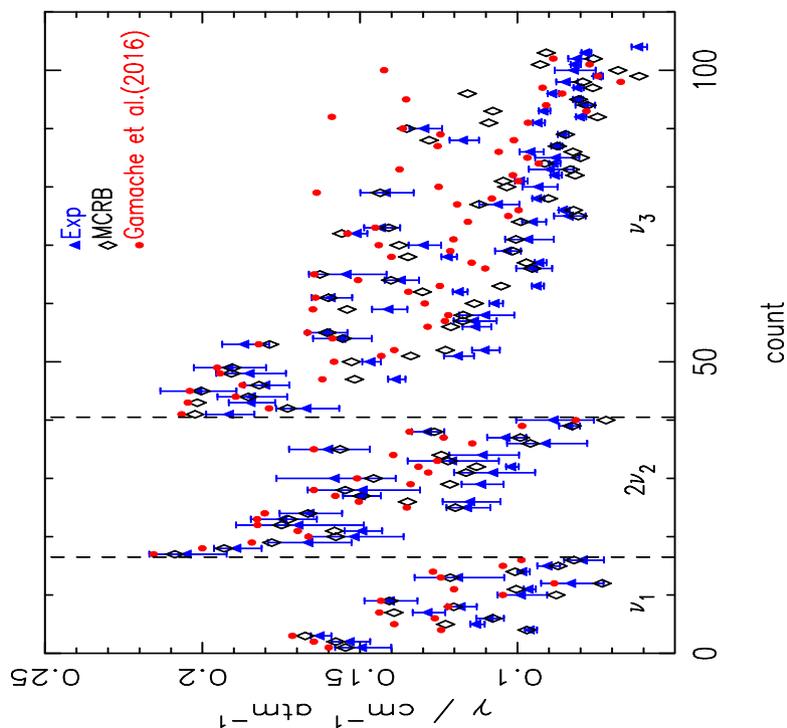


Fig. 5. CO₂-broadened half-widths (cm⁻¹ atm⁻¹) of H₂¹⁶O transitions in the 2.7 μm region versus line count. For each vibrational band the points are ordered by the upper and lower state energy ordered index; index = $J'(J'+1) + Ka - Kc + 1$. Shown are measurements (Exp) from this work (blue solid triangle symbols), MCRB calculations from this work (black open diamond symbols), and the previous calculations of Gamache et al. [17] (red solid circle symbols). The data are arranged such that the first 16 points are ν_1 band data, points 17–40 are $2\nu_2$ band data, and points 41–105 are ν_3 band data. The error bars given for the measurements correspond to the 2σ statistical uncertainty provided by the fit. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

Table 4

Comparison of the measured half-widths and line shifts with the MCRB calculations of this work and with the calculations of Gamache et al. [17] in the spectral region around 2.7 μm. Given are the average percent difference (APD) and standard deviation (SD) in percent for the half-widths and the average deviation (Adev) and standard deviation in cm⁻¹ atm⁻¹ for the line shifts.

	Band	1 0 0 ← 0 0 0	0 2 0 ← 0 0 0	0 0 1 ← 0 0 0
	# points	16	24	66
Half-width				
Exp-MCRB	APD	-1.1	-1.8	-3.1
	SD	5.7	4.9	6.0
Exp-Gamache et al. [32].	APD	-13.8	-12.0	-11.8
	SD	9.7	8.5	9.4
Line shift				
Exp-MCRB	Adev [†]	-0.0037	-0.0014	-0.0025
	SD [†]	0.0056	0.0062	0.0060
Exp-Gamache et al. [32].	Adev [†]	-0.0750	-0.0666	-0.0848
	SD [†]	0.0623	0.0683	0.0664

[†] in units of cm⁻¹ atm⁻¹.

The standard deviations show similar features. Given the relatively large value of the quadrupole moment of CO₂, it could be argued that the vibrational dependence of the half-widths should not be too large. However, looking at the calculations to adjust the potential, the average difference between the potential used in Ref. [17] and the results from the final potential of this work (pot 23) is roughly 9%. This result suggests that the vibrational dependence for these bands is ~3–4%. We will study the vibrational dependence of the half-width in a future work.

Fig. 6 is a plot of the measured line shifts with error bars (blue solid triangle symbols), our MCRB calculations (black open diamond symbols), and the data of Gamache et al. [17]. Because the study of Gamache et al. [17] only made calculations for the rotation band, the values of the shifts in the database are from various approximate schemes from the work of Jacquemart et al. [39], or scaled directly from δ_{air} from HITRAN2012 [40] ($\delta_{\text{CO}_2} = \delta_{\text{air}} * 4.211$).

In the plot, the red open star symbols correspond to Jacquemart et al.'s Section 4.2.3 approximation; the red open circle symbols correspond to Jacquemart et al.'s Section 4.1 approximation; the red open plus symbols correspond to Jacquemart et al.'s Section 4.2.1 approximation; and the red asterisk symbols correspond to scaling the air-induced pressure shift. The statistics for the comparison of the line shifts are in the bottom part of Table 4. The average deviation between the measured and calculated line shifts goes between -0.001 to -0.004 cm⁻¹ atm⁻¹ with standard deviations around 0.006 cm⁻¹ atm⁻¹ for the 3 vibrational bands studied. The comparison with the approximate schemes used in Ref. [17] shows much larger average deviations, -0.07 to -0.08 cm⁻¹ atm⁻¹, and standard deviations, ~0.065 cm⁻¹ atm⁻¹. This figure and the statistics demonstrate the need for primary measurement or calculation to obtain good line shifts. It has been shown before [11,41,42] that scaling techniques lead to very large errors, and it

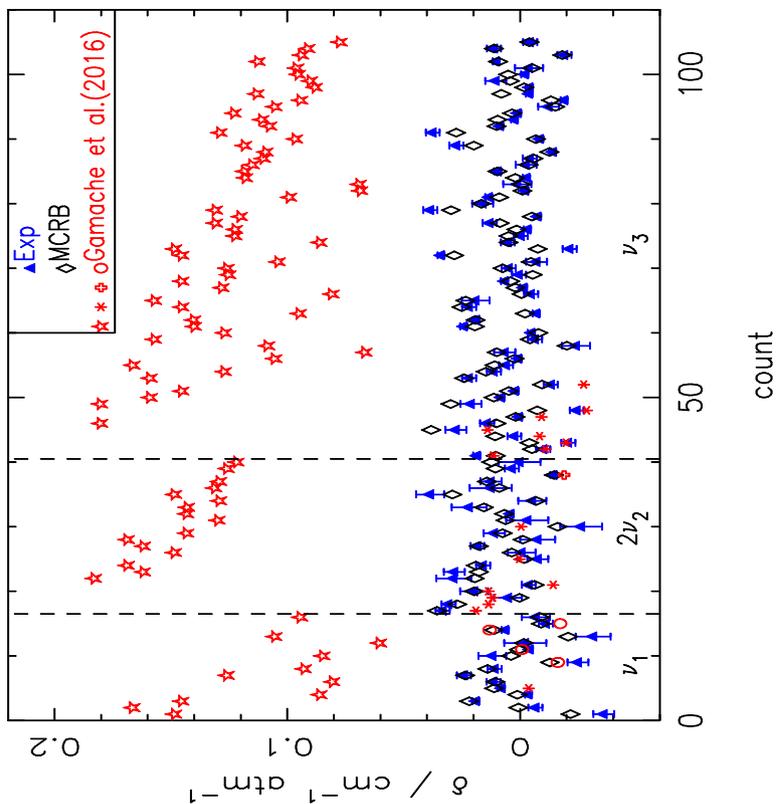


Fig. 6. CO₂ pressure-induced line shifts (cm⁻¹ atm⁻¹) of H₂¹⁶O transitions in the 2.7 μm region versus line count. For each vibrational band the points are ordered by the upper and lower state energy ordered index; index = J'(J + 1) + Ka - Kc + 1. Shown are measurements (Exp) from this work (blue solid triangle symbols), MCRB calculations from this work (black open diamond symbols), from Gamache et al. [17]: the red open star symbols correspond to Jacquemart et al.'s [39] Section 4.2.3 approximation; the red open circle symbols correspond to Jacquemart et al.'s Section 4.1 approximation; the red open plus symbols correspond to Jacquemart et al.'s Section 4.2.1 approximation; and the red asterisk symbols correspond to scaling the air-induced pressure shift. See text for details. The data are arranged such that the first 16 points are ν₁ band data, points 17–40 are 2ν₂ band data, and points 41–105 are ν₃ band data. The error bars given for the measurements correspond to the 2σ statistical uncertainty provided by the fit. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

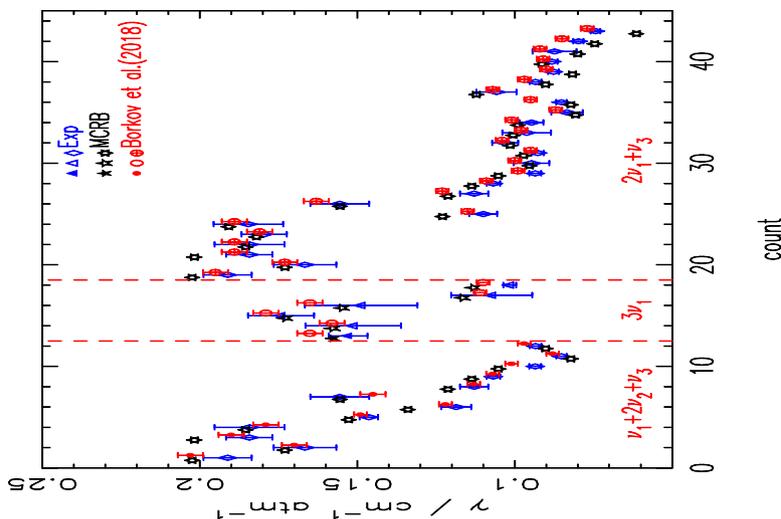


Fig. 7. Comparison of the half-widths (cm⁻¹ atm⁻¹) from the measurements (Exp) blue solid triangle, open triangle, and open diamond symbols for our ν₁, 2ν₂, and ν₃ band data, respectively), the MCRB calculations (black solid star, open star, and star of David symbols for our ν₁, 2ν₂, and ν₃ band data, respectively), and the Borkov et al. [43] data (red solid circle, open circle, and open circle with a cross symbols for our ν₁, 2ν₂, and ν₃ band data, respectively) versus line count. The data are arranged such that the first 13 points are ν₁ + 2ν₂ + ν₃ band data, points 14–19 are 3ν₁ band data, and points 20–45 are 2ν₁ + ν₃ band data. For each vibrational band the points are ordered by the upper and lower state energy ordered index; index = J'(J + 1) + Ka - Kc + 1. The error bars given for the measurements correspond to the 2σ statistical uncertainty provided by the fit. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

was stated [17] that the error from the semi-empirical line shift prediction routine can be large.

The line lists for the two previous spectral regions are given in the supplementary data.

5.3. Comparisons with data around 0.95 μm

Borkov et al. [43] have measured the broadening and shift parameters of the water vapor spectral lines in the 10,100–10,800 cm^{-1} region induced by pressure of carbon dioxide. We compared the measurements and calculations made here to their results for the half-width. Note, because the line shift is strongly dependent on vibration, a comparison of our line shifts to their data was not done. Borkov et al. made measurements for transitions in the $\nu_1 + 2\nu_2 + \nu_3$, $3\nu_1$, and $2\nu_1 + \nu_3$ bands. These data are compared with our measurements and calculations in the 2.7 μm region (ν_1 , $2\nu_2$, and ν_3 bands). Fig. 7 shows the half-widths in $\text{cm}^{-1} \text{atm}^{-1}$ for the our measurements with error bars (blue solid triangle, open triangle, and open diamond symbols for our ν_1 , $2\nu_2$, and ν_3 band data, respectively), the MCRB calculations shifted slightly to the left (black solid star, open star, and star of David symbols for our ν_1 , $2\nu_2$, and ν_3 band data, respectively), and the Borkov et al. data with error bars shifted slightly to the right (red solid circle, open circle, and open circle with a cross symbols for their $\nu_1 + 2\nu_2 + \nu_3$, ν_1 , $3\nu_1$, and $2\nu_1 + \nu_3$ band data, respectively) versus line count. The data are arranged such that the first 13 points are $\nu_1 + 2\nu_2 + \nu_3$ band data, points 14–19 are $3\nu_1$ band data, and points 20–45 are $2\nu_1 + \nu_3$ band data (labeled at the bottom of the figure). In total 45 transitions were compared between the data sets. The statistics for the comparison of our measurements with the data of Borkov et al. are an APD of 1.5% with a SD of 7.3%; the comparison of our calculations to the data of Borkov et al. are an APD of -3.9% with a SD of 3.4%. The statistics imply a constant shift of our calculations, within 3–4%, from the Borkov et al. data. Even though the spectral regions of the two data sets are separated by about 6000 cm^{-1} , there is good agreement among the data sets. This result implies the vibrational dependence is of order $\sim 3\%$, in agreement to the results from the previous section.

6. Conclusion

New laboratory measurements of line-shape parameters of H_2O broadened by CO_2 were performed. Two spectral regions were probed, i.e. 2.7 and 6 μm , as they are of interest for the planetary atmospheres remote sensing. MCRB calculations of the half-width, line shift, and their temperature dependence were made for the cold bands of H_2O in the spectral range 1300–5000 cm^{-1} . The results were compared with the literature values [11,17,43]. There is good agreement between the measured and calculated half-widths and line shifts with the data of Brown et al. in the 6 μm region. In the 2.7 μm region, our measurements and MCRB calculations agree well for both the half-widths and line shifts. Comparison with the half-widths and line shifts from Ref. [17] are not as good, showing the effects of neglecting vibrational dependence for the half-widths and using approximate prediction routines for the line shifts. Comparisons of our measured and calculated half-widths with the data of Borkov et al. [43] on the 0.9 μm region show good agreement. The measurements are available as supplementary files. The calculations in the range 1300–5000 cm^{-1} are available as a supplementary file. In the future, the MARS database [17] will be updated with the new parameters from this work and will be made publicly available, with a read_me.txt file, as a file from one of the authors website (faculty.uml.edu/robert_gamache). Note, using the improved model of temperature dependence [38], the same database can be used for Mars, Venus, or any other planetary atmosphere rich in CO_2 .

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Supplementary materials

Supplementary material associated with this article can be found, in the online version, at doi:10.1016/j.jqsrt.2019.04.012.

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