

# Geophysical Research Letters

## RESEARCH LETTER

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### Key Points:

- Increase of hydrogen chloride (HCl) during the southern summer is annually repeated. It shows that HCl formation is independent from a global dust storm
- Vertical distributions of HCl are strikingly similar to water vapor. It suggests that uptake by water ice clouds plays an important role
- A rapid decrease of HCl at the end of the southern summer suggests the presence of a strong sink in addition to the photochemical loss

### Supporting Information:

Supporting Information may be found in the online version of this article.

### Correspondence to:



























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## Annual Appearance of Hydrogen Chloride on Mars and a Striking Similarity With the Water Vapor Vertical Distribution Observed by TGO/NOMAD

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**Abstract** Hydrogen chloride (HCl) was recently discovered in the atmosphere of Mars by two spectrometers onboard the ExoMars Trace Gas Orbiter. The reported detection made in Martian Year 34 was transient, present several months after the global dust storm during the southern summer season. Here, we present the full data set of vertically resolved HCl detections obtained by the NOMAD instrument, which covers also Martian year 35. We show that the particular increase of HCl abundances in the southern summer season is annually repeated, and that the formation of HCl is independent from a global dust storm event. We also find that the vertical distribution of HCl is strikingly similar to that of water vapor, which suggests that the uptake by water ice clouds plays an important role. The observed rapid decrease of HCl abundances at the end of the southern summer would require a strong sink independent of photochemical loss.

**Plain Language Summary** A new species, hydrogen chloride (HCl), was recently discovered in the atmosphere Mars. Whereas, this gas plays a key role in the atmospheric chemistry on Earth and Venus, the chlorine cycle on Mars is not understood. HCl was found just after the global dust storm, so a connection between the storm and appearance of HCl was suggested. This study shows that HCl is detected even in a year without a global dust storm, which demonstrates that the formation of HCl is primary independent from global dust storms. This study also finds that the vertical distributions of HCl and water vapor are strikingly similar in shape. As the vertical profile of water vapor is controlled by the formation of water ice clouds, it suggests that HCl is taken up by water ice clouds. Finally, we observe that HCl is rapidly disappearing at the end of the southern summer on Mars. We demonstrate that this fast destruction cannot be explained by the standard photochemistry, which suggests that another (unknown) process plays a key role.

## 1. Introduction

Chlorine chemistry plays an important role in the atmosphere of Earth and Venus. On Earth, small amounts of chlorine can have a dramatic impact on stratospheric ozone (e.g., Solomon, 1999). On Venus, chlorine is involved in the main photochemical cycles (e.g., Mahieux et al., 2015; Sandor and Clancy, 2018). In contrast,

on Mars, chlorine photochemistry was considered to be negligible because of the low upper limits provided by the previous measurements (Hartogh et al., 2010; Krasnopolsky et al., 1997; Villanueva et al., 2013). However, the first detection of hydrogen chloride (HCl) in the atmosphere of Mars was made by the Atmospheric Chemistry Suite (ACS) and also confirmed by the Nadir and Occultation for Mars Discovery (NOMAD) onboard the ESA-Roskosmos ExoMars Trace Gas Orbiter (TGO) (Korablev et al., 2021). This discovery of HCl is the first detection of any chlorine species in the Martian atmosphere. It demonstrates an active presence of chlorine photochemistry and HCl is expected to be the main reservoir of chlorine species in the atmosphere of Mars (Lefèvre and Krasnopolsky, 2017). Perchlorate salts  $\text{ClO}_4^-$  (Glavin et al., 2013; Hecht et al., 2009) and chloride-bearing minerals  $\text{Cl}^-$  (Osterloo et al., 2008) are known to be widely present at the Mars surface, thus an interaction between surface and atmosphere may support chlorine photochemistry. Model and laboratory studies investigated the potential role of atmospheric photochemical processes in the formation of perchlorates (e.g., Catling et al., 2010; Smith et al., 2014; Wilson et al., 2016; Wu et al., 2018, and references therein). However, the chlorine cycle on Mars is still not understood. The discovery made by Korablev et al. (2021) hints at the transient nature of HCl: the detections are mainly confined after the global dust storm in the southern summer season. The observed fast temporal variation can be a key to identify the formation and destruction mechanisms of HCl. The discovery was based on the data taken in Martian year (MY) 34. This study presents the results of the extended NOMAD data set that also includes almost the full MY 35. We investigate the vertical distribution of HCl volume mixing ratios (vmr) and their similarities with simultaneously obtained water vapor and aerosol profiles. An analysis of the Cl isotopic ratio in HCl with the NOMAD data are presented in an accompanying paper (Liuzzi et al., 2021).

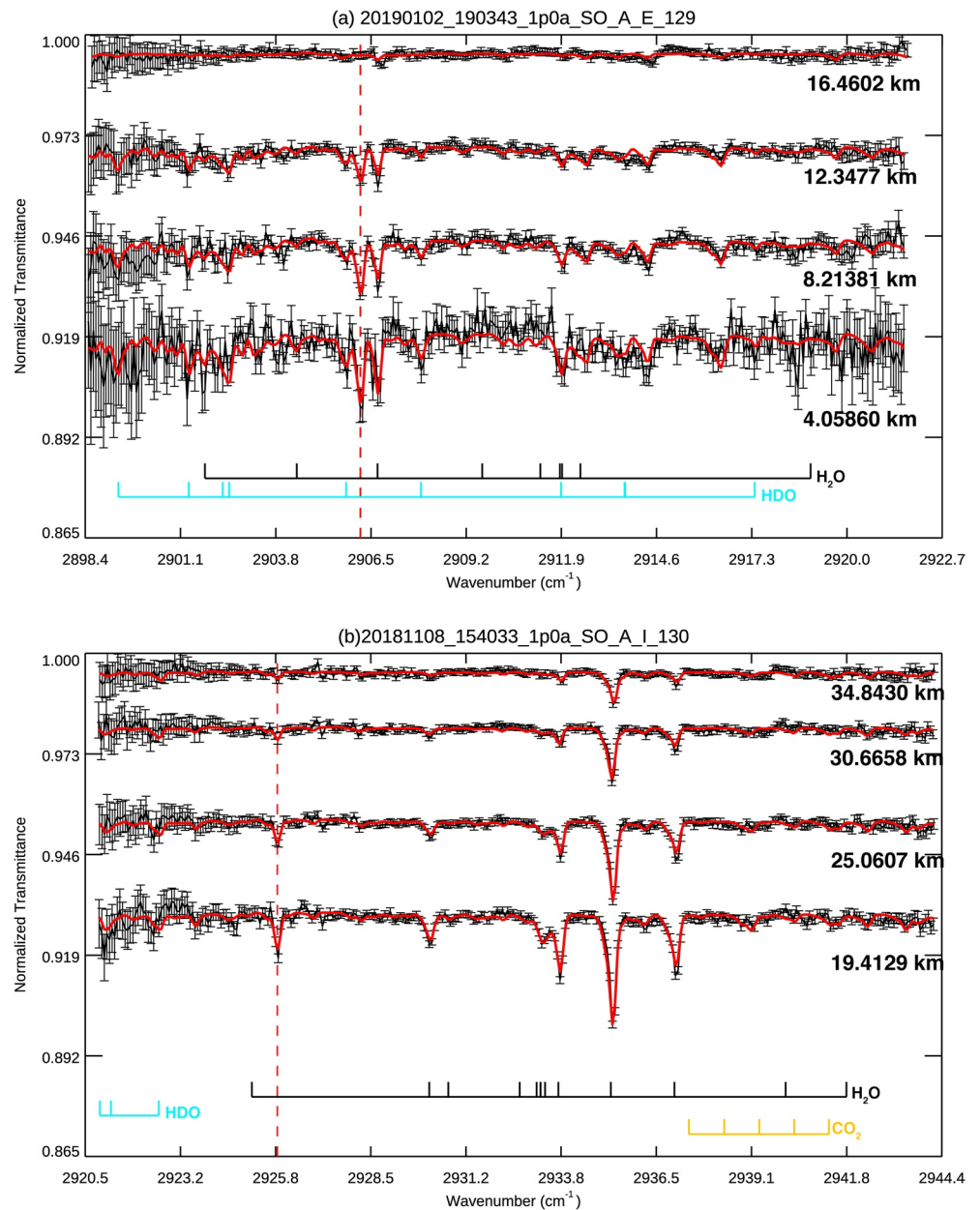
## 2. Methods

### 2.1. Instrument—NOMAD Onboard TGO

NOMAD, a spectrometer operating in the spectral ranges between 0.2 and 4.3  $\mu\text{m}$  onboard ExoMars TGO (Vandaele et al., 2018), has three spectral channels: a solar occultation channel (SO—Solar Occultation; 2.3–4.3  $\mu\text{m}$ ), a second infrared channel capable of nadir, solar occultation, and limb sounding (LNO—Limb Nadir and solar Occultation; 2.3–3.8  $\mu\text{m}$ ), and an ultraviolet/visible channel (UVIS—Ultraviolet and Visible Spectrometer, 200–650 nm). The infrared channels (SO and LNO) have relatively high spectral resolution ( $\lambda/d\lambda \sim 10,000$ – $20,000$ ) provided by an echelle grating used in combination with an Acousto Optical Tunable Filter (AOTF) which selects diffraction orders (Neefs et al., 2015). The sampling rate for the solar occultation measurement is 1 s, which provides high vertical sampling ( $\sim 1$  km) with high vertical resolution ( $\sim 2$  km) from near the surface to 200 km altitude. Thanks to the instantaneous change of the observing diffraction orders achieved by the AOTF, the SO channel is able to measure five or six different diffraction orders per second in solar occultation mode. One of the most remarkable capabilities of NOMAD is its high spectral resolution in the infrared range. This supports investigations of: (a) vertical profiles for atmospheric constituents such as carbon dioxide, carbon monoxide, water vapor, and their isotopic ratios (Aoki et al., 2019; Vandaele et al., 2019; Villanueva et al., 2021); and (b) sensitive searches for organic species (such as  $\text{CH}_4$ ,  $\text{C}_2\text{H}_4$ ,  $\text{C}_2\text{H}_6$ ,  $\text{H}_2\text{CO}$ ; Korablev et al., 2019; Knutsen et al., 2021) and other trace gases (such as HCl, HCN,  $\text{HO}_2$ ,  $\text{H}_2\text{S}$ ,  $\text{N}_2\text{O}$ , and OCS) by solar occultation measurements with the SO channel. In this study, we analyze the solar occultation measurements acquired by the NOMAD SO channel during the period from April 22, 2018 to December 18, 2020 (from a solar longitude ( $L_s$ ) of  $163^\circ$  in MY 34 to  $L_s = 333^\circ$  in MY 35), which covers southern summer periods in MY 34 and 35. For this study, a total of 732 occultations that include diffraction order 129 ( $2,889$ – $2,921$   $\text{cm}^{-1}$ ) or 130 ( $2,920$ – $2,943$   $\text{cm}^{-1}$ ) have been processed. These two diffraction orders contain a strong HCl line (see Figure 1) and are normally acquired together with diffraction order 134 ( $3,010$ – $3,034$   $\text{cm}^{-1}$ ), order 136 ( $3,056$ – $3,080$   $\text{cm}^{-1}$ ), or order 145 ( $3,258$ – $3,284$   $\text{cm}^{-1}$ ), which include strong water lines in support of direct comparisons between HCl and water vapor vertical profiles.

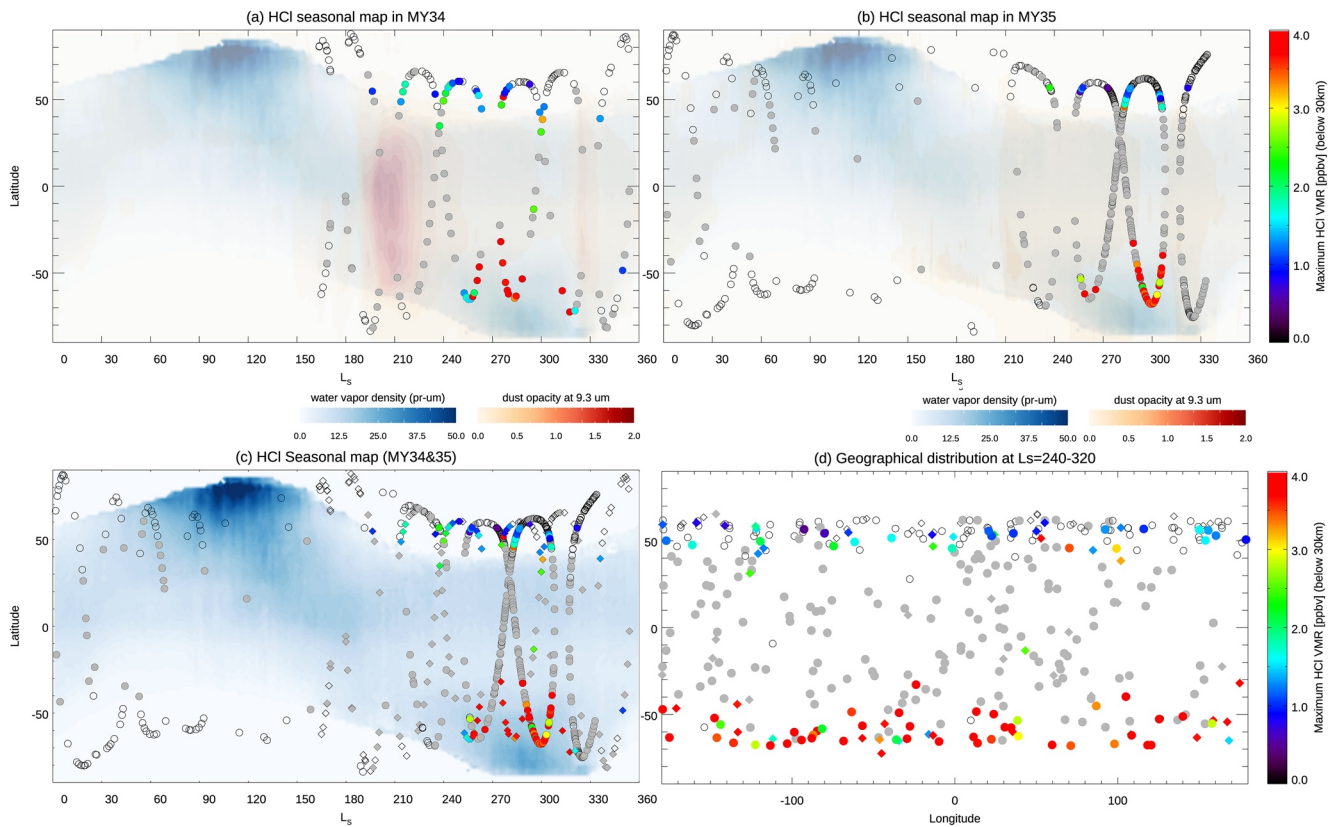
### 2.2. Retrieval of HCl and Water Vapor Volume Mixing Ratio

The reduction and analysis of the NOMAD data are basically conducted with the same methodology as the previous water vapor study (Aoki et al., 2019; Vandaele et al., 2019). The retrievals of HCl and  $\text{H}_2\text{O}$  are performed with the ASIMUT-ALVL radiative transfer and inversion code (Vandaele et al., 2006).  $\text{H}_2\text{O}$ , HDO,  $\text{CO}_2$ , and HCl molecular absorptions are taken into account in the radiative transfer calculation and



**Figure 1.** Examples of the NOMAD-SO spectra with signatures of HCl: (a) measurements in the diffraction order 129 performed on January 2, 2019 ( $L_s = 317.4^\circ$ , Latitude =  $72^\circ\text{S}$ , Longitude =  $46^\circ\text{W}$ ); (b) measurements in the diffraction order 130 performed on 8 November 2018 ( $L_s = 284.5^\circ$ , Latitude =  $63^\circ\text{S}$ , Longitude =  $149^\circ\text{E}$ ). The best-fit synthetic spectra are shown by the red curves. The vertical dashed lines in red represent the position of the HCl lines. The other spectral signatures are absorptions due to water vapor and  $\text{CO}_2$ . The bottom markers show the position of the strong  $\text{H}_2\text{O}$  ( $S > 1 \times 10^{-24}$  cm, black), HDO ( $S > 1 \times 10^{-24}$  cm, light blue), and  $\text{CO}_2$  ( $S > 1 \times 10^{-28}$  cm, orange) lines in the main diffraction order.

the absorptions are calculated using the following spectroscopic database: HITRAN 2016 database (Gordon et al. 2017) for  $\text{CO}_2$  and HCl, and the water line list for  $\text{CO}_2$ -rich atmospheres by Gamache et al. (2016) for  $\text{H}_2\text{O}$  and HDO. The temperature, pressure, and  $\text{CO}_2$  vmr of the simulated atmosphere is obtained from the Global Environmental Multiscale Mars model (GEM-Mars) (Daerden et al., 2019; Neary & Daerden, 2018) which takes into account the effects of the global dust storm in MY34 (Neary et al., 2020). The HCl and  $\text{H}_2\text{O}$  retrievals are performed using the Optimal Estimation Method (OEM) (Rodgers, 2000) for each spectrum at each tangential altitude independently. The retrieved HCl and  $\text{H}_2\text{O}$  abundances are finally averaged



**Figure 2.** (a–c) Solar longitude—Latitude map of the maximum HCl mixing ratio (ppbv) below 30 km in (a) MY 34, (b) MY 35, and (c) MY34–35 plotted together. Only 3- $\sigma$  detections are shown here. The gray points show observations corresponding to dust top altitude greater than 25 km, for which it is generally difficult to perform robust retrievals. The background color maps show the column-integrated water vapor density (pr- $\mu\text{m}$ ) obtained by MGS/TES for Mars year 26 (Smith, 2002, 2006) and the column-integrated dust opacity at 9.3  $\mu\text{m}$  normalized for 610 Pa for MY 31 (b) and MY 34 (a) (Montabone et al., 2020). (d) Geographical distribution of HCl below 30 km in the seasonal range between  $L_s = 240^\circ$  and  $320^\circ$ .

with an interval of 2 km to obtain the vertical profiles. The averages are weighted by their standard errors (weighted average). In addition to the HCl and  $\text{H}_2\text{O}$  abundances, the vertical profiles of aerosols extinction are investigated. The extinction profiles are simply derived from the averaged transmittances (at diffraction order 129 or 130) based on the onion-peeling method (Wilquet et al., 2012). Figure 1 shows an example of the NOMAD spectra at diffraction order 129 (Figure 1a) and order 130 (Figure 1b). Clear signatures of HCl are identified at  $2,906.24\text{ cm}^{-1}$  in order 129 and  $2,925.89\text{ cm}^{-1}$  in order 130. HCl is usually detected in the lower atmosphere (<20–25 km) but sometimes up to 35–40 km altitude as shown in Figure 1b.

### 3. Results

#### 3.1. Seasonal, Spatial, and Interannual Variations of HCl

Figures 2a–2c show the season-latitude maps of the maximum HCl vmr below 30 km observed by NOMAD in MY 34 (Figure 2a), MY 35 (Figure 2b), and plotted together (Figure 2c). As shown in Korablev et al. (2021), a few ppbv of transient HCl was detected in the Mars atmosphere. In MY 34, two strong dust storms occurred on Mars—the global dust storm at  $L_s = 190^\circ$ – $250^\circ$  and the regional dust storm at  $L_s = 320^\circ$ – $340^\circ$  (e.g., Aoki et al., 2019; Liuzzi et al., 2020). Almost all of the HCl detections were made in the period between the two dust storms ( $L_s = 240^\circ$ – $320^\circ$ ) (Figure 2a). The period of HCl detections also correspond to the southern summer season on Mars. The temporal variation of the HCl detections in MY 35 is similar to that in MY 34: most of the HCl detections are made in the southern summer period ( $L_s = 240^\circ$ – $320^\circ$ ) (Figure 2b). We use the dust climatology of MY 31 as a proxy, as it looks similar to that of MY 35. Since there is no global dust storm in MY35, it is demonstrated that the formation of HCl is independent from a global dust storm event. In both MY 34 and 35, higher abundances of HCl are observed in the southern hemisphere (local

summer,  $L_s = 270^\circ$ ), particularly at high latitudes, which corresponds in time and location to the water vapor enhancement caused by the south polar cap sublimation (Figure 2c). Because the observations before  $L_s \sim 240^\circ$  were not possible in the southern hemisphere due to dust obscuration, it is unclear whether this represents an increase in HCl, or if high abundances were already present. In contrast, the vmr of HCl is below the detection limit ( $\sim 0.3$  ppbv) at the northern hemisphere summer solstice ( $L_s = 90^\circ$ ), even at high latitudes where the column integrated abundances of water vapor are large.

Figure 2d shows the geographical distribution of the maximum HCl vmr below 30 km observed by NOMAD in the southern summer period ( $L_s = 240^\circ\text{--}320^\circ$ ) of MY 34 and 35. The detections are dispersed over all observed longitudes, such that HCl is not confined to a specific longitude region. Most of the detections are performed at latitudes higher than  $45^\circ$ . This is because higher dust content in the atmosphere at lower latitudes obscures solar occultation viewing. In other words, the non-detection of HCl at low-latitudes means that the abundance of HCl cannot be measured there. There is a clear north-south asymmetry—the maximum vmr of HCl is larger in the southern hemisphere than in the northern hemisphere.

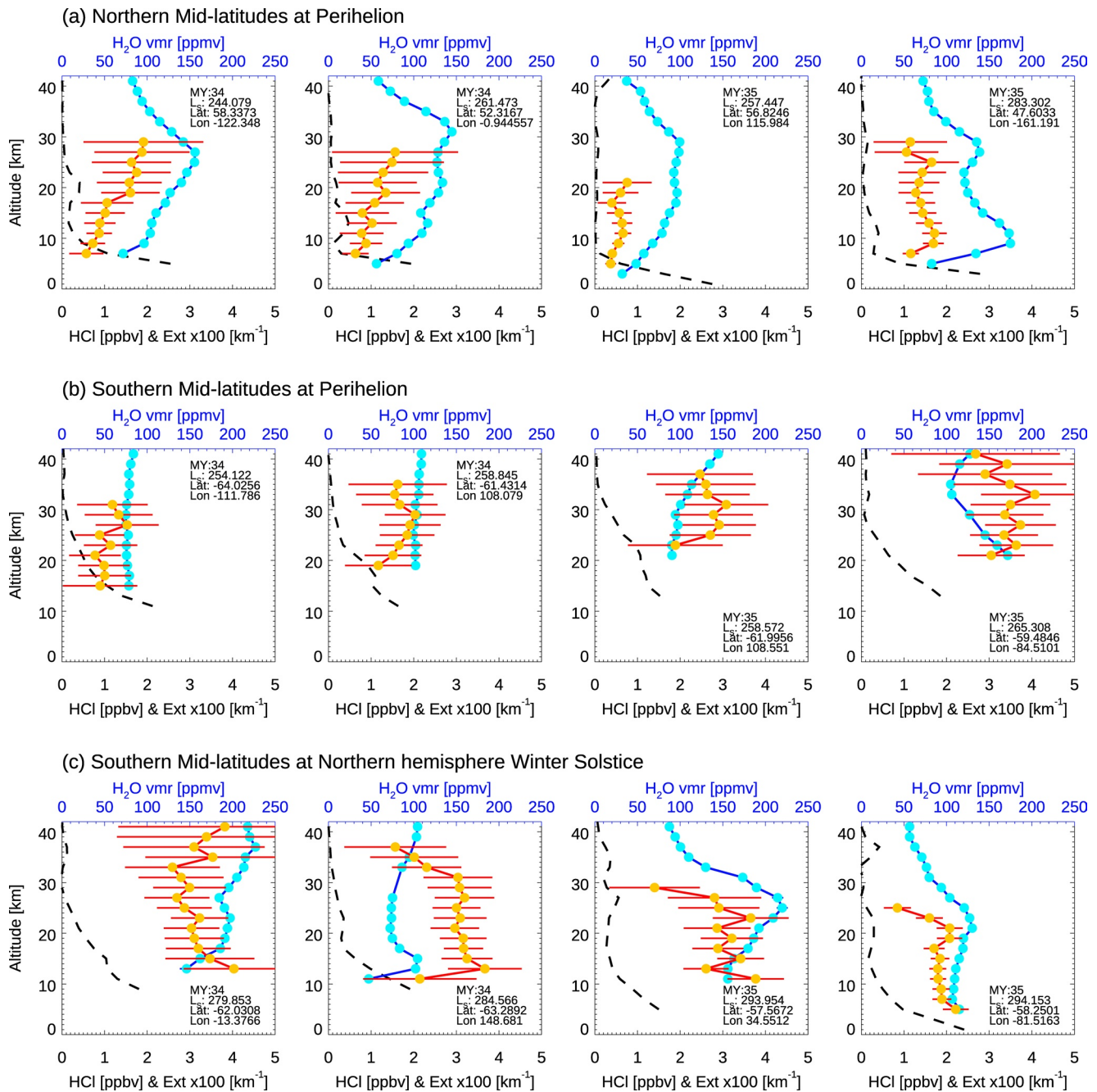
### 3.2. Vertical Distribution of Hydrogen Chloride, Water Vapor, and Aerosols

Figure 3 shows several representative vertical profiles of HCl, water vapor, and aerosol extinction observed during the same occultation. The seasonal variations of the HCl vertical distributions measured in MY 34 and 35 are quite similar. At northern mid-latitudes ( $50^\circ\text{--}60^\circ\text{N}$ ) around perihelion ( $L_s = 240^\circ\text{--}270^\circ$ ) (Figure 3a), water vapor vmr increased toward the higher altitudes with a maximum around 30 km. This decrease below 30 km reflects water vapor loss at lower altitudes due to the condensation over the seasonal polar cap (e.g., Daerden et al., 2019). In MY 34, because of increased upper level transport associated with the global dust storm, water is also enhanced at higher altitudes (Aoki et al., 2019; Fedorova et al., 2020; Neary et al., 2020). HCl also shows a decrease toward lower altitudes. At the southern mid-to-high latitudes ( $50^\circ\text{--}70^\circ\text{S}$ ) during the same period (Figure 3b), water vapor vmr are constant throughout the atmosphere and HCl follows this same uniform distribution.

This constant mixing continues after perihelion ( $L_s = 270^\circ\text{--}300^\circ$ ), although in MY 35, water and HCl vmr start to decrease between 20 and 30 km (Figure 3c). Note that in this seasonal range at  $L_s = 270^\circ\text{--}300^\circ$ , we observe larger water vapor mixing ratios (because of the sublimation of the south polar cap) as well as the largest observed HCl vmr (4 ppbv) both in MY 34 and 35. Moreover, HCl is detected at the highest altitudes (35–40 km). Water vapor is still abundant in the later period ( $L_s = 300^\circ\text{--}330^\circ$ ) but rapidly drops above the hygropause, which is located around 15–20 km (Figure 3d). Interestingly, HCl also dramatically drops at the same altitudes. Finally, in the northern hemisphere during the summer solstice ( $L_s = 90^\circ\text{--}120^\circ$ ) (Figure 3e), HCl is not detected anymore while a large amount of water vapor is observed at lower altitudes (below 10 km) because of the sublimation of the northern polar cap. This is a low-dust season, and the upper limit of HCl below 10 km is  $\sim 0.3$  ppbv, which is well below the abundances detected in the southern summer season.

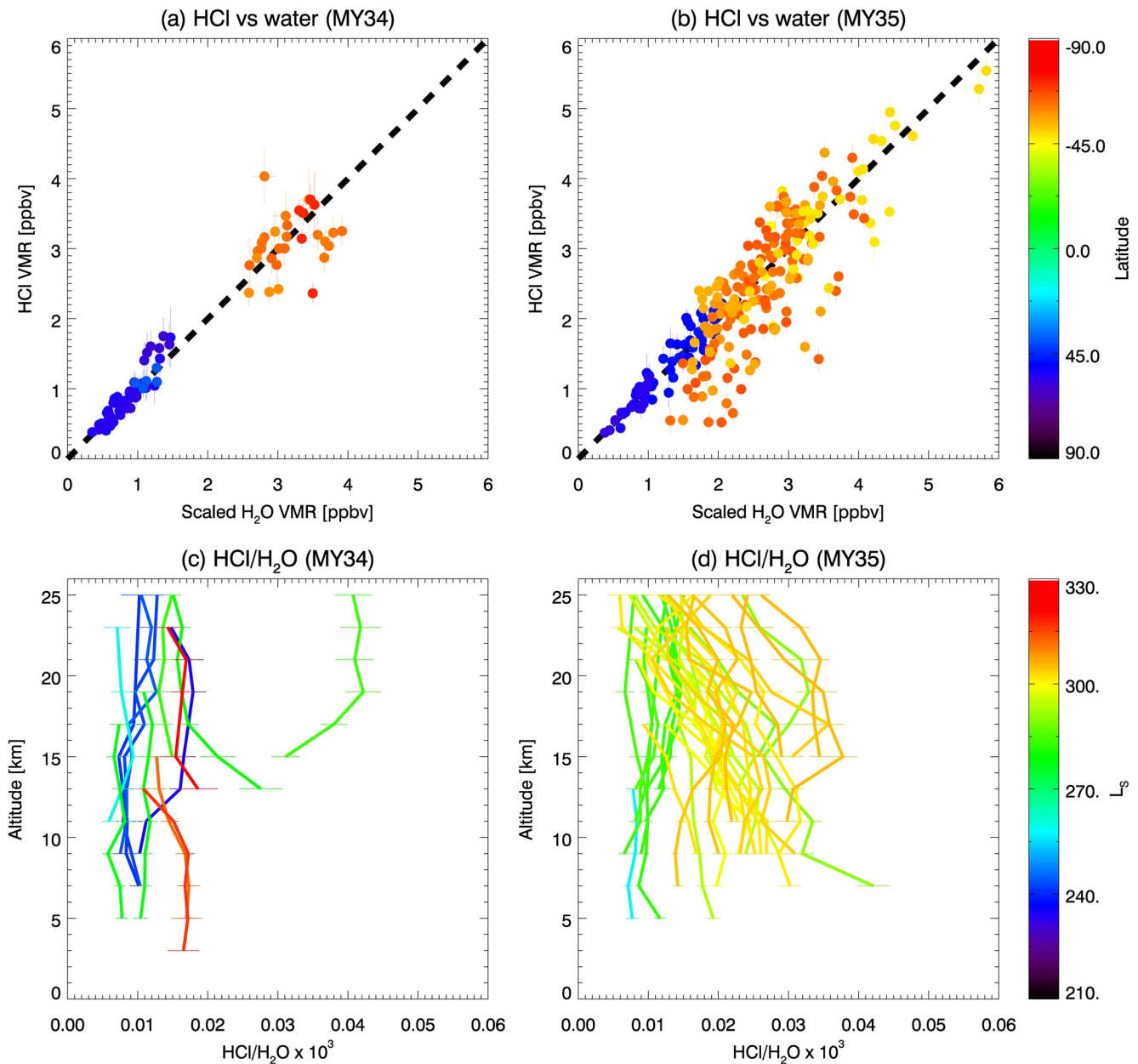
Such a temporal variation of the HCl vertical profiles observed by NOMAD is consistent with the simultaneous measurements by ACS reported earlier (Korablev et al., 2021). In this study, we emphasize that the vertical structures of HCl and water vapor are strikingly similar. Figures 4a and 4b shows the HCl vmr against scaled water vapor vmr for all the HCl detections below 25 km. The water vapor mixing ratios in this figure were scaled for each profile by the ratio of the partial column densities of HCl and water vapor below 25 km (see supporting information Section S1, for more details). This scaling process comes down to shifting the upper and lower x-axes in Figure 3 such that the lower part of the HCl and  $\text{H}_2\text{O}$  profiles are overlapping, and so allows us to evaluate the similarity in the shapes of the vertical distributions. The partial column densities were calculated using collocated GEM-Mars temperature and pressure profiles (Daerden et al., 2019; Neary et al., 2020) for the altitudes where HCl was detected. It was preferred to work with partial column densities rather than the ratio of the vmr at one specific altitude level, because there is no unique altitude level at which HCl was detected in all the profiles.

Figures 4a and 4b demonstrates a strong similarity of the shapes of HCl and water vapor vertical structures in the perihelion season. The water vapor profile is shaped by the formation of clouds and their subsequent sedimentation over the course of one or many consecutive sols, counteracting the ascend of water



**Figure 3.** Representative vertical profiles of HCl, water vapor, and aerosols extinction at (a) the northern mid-latitudes ( $50^{\circ}$ – $60^{\circ}$ N) at  $L_s = 240^{\circ}$ – $270^{\circ}$ ; (b) the southern high-latitudes ( $60^{\circ}$ – $70^{\circ}$ S) during the perihelion ( $L_s = 240^{\circ}$ – $270^{\circ}$ ); (c) the southern mid-latitudes ( $30^{\circ}$ S and  $60^{\circ}$ S) at  $L_s = 270^{\circ}$ – $300^{\circ}$ ; (d) the southern high-latitudes ( $60^{\circ}$ S– $70^{\circ}$ S) at  $L_s = 300^{\circ}$ – $330^{\circ}$ ; and (e) the northern mid-latitudes ( $50^{\circ}$ – $60^{\circ}$ N) at  $L_s = 90^{\circ}$ – $120^{\circ}$ . The red-orange, blue-light blue, and dashed black curves show HCl volume mixing ratio (0–5 ppbv), water vapor mixing ratio (0–250 ppmv), and extinction coefficient (0.00–0.05  $\text{km}^{-1}$ ), respectively. The error bars represent 3- $\sigma$  uncertainty.

vapor with the normal circulation (Neary et al., 2020). This causes a decrease of the water vapor vmr at the “hygropause” level (e.g., Daerden et al., 2019, their Figure 13). In polar winter conditions, the water vapor profile is also decreasing toward the surface due to strong surface deposition. We see both types of profiles in Figure 3. The similarity of the HCl profiles to those of water vapor suggests that HCl is depositing with water vapor, both at the hygropause and at the surface in polar winter. Laboratory work for temperatures (194–228 K) and pressures (133 Pa) representative of Mars (Kippenberger et al., 2019), and for similar HCl



**Figure 4.** (a and b) HCl volume mixing ratios are plotted against scaled water vapor mixing ratios, for all profiles combined (a: MY34, b: MY35). The water vapor mixing ratios are scaled by the ratio of the HCl and H<sub>2</sub>O partial column densities below 25 km within each profile (see supporting information S1). When HCl and H<sub>2</sub>O profiles are similar in shape, their points will fall on or near the identity line. The color scale represents the latitude of the observations (c and d) HCl/H<sub>2</sub>O ratio for all profiles. The color scale represents solar longitude ( $L_s$ ) of the observations.

partial pressures as found here, demonstrated an extensive, continuous uptake of HCl in water ice during growth of ice particles. This seems to indicate that HCl, once introduced in the atmosphere, tracks with water vapor. However, this behavior does not address the ultimate source or loss processes for Mars atmospheric HCl, as will be discussed below. When the ratio HCl/H<sub>2</sub>O is plotted for all the profiles combined, it is not decreasing over time and altitude (Figures 4c and 4d). This could imply that the uptake of HCl on water ice clouds is not a permanent sink and that HCl is released again when water ice sublimates. However, it can also mean that new chlorine-bearing species are formed at the water ice cloud particles, that can become a new source of HCl (see Discussion for more details).

## 4. Discussion

### 4.1. Source of HCl

The first detections of HCl were obtained mainly just after the MY 34 global dust storm. Therefore, Korablev et al. (2021) proposed that dust—including chloride-bearing materials—lifted to high altitudes could release chlorine into the atmosphere by UV irradiation. This would ultimately lead to the formation of HCl through the fast reaction of chlorine with hydroperoxyl that results from photolysis of water vapor also lifted to high altitudes (Lefèvre and Krasnopolsky, 2017; Wilson et al., 2016):



However, the repeated detections of equivalent HCl abundances in the same season in MY 35 clearly indicates that a global dust storm is not required, the annually enhanced dust abundances in the perihelion season appearing to be sufficient to produce HCl.

The strong enhancement of HCl at the southern high latitudes in the southern summer season suggests that the main source of atmospheric chlorine could be there. We note that further measurements are needed to confirm the source as a large part of the NOMAD observations at low- and mid-latitudes could not detect HCl because of strong dust presence. Nevertheless, if the source is the southern high latitudes, the produced HCl can be transported to the northern hemisphere by the global circulation. Theoretical studies show that the time scale of the global mixing from 60°S latitude to north polar latitudes can be as low as 20 sols or less in this season (Waugh et al., 2019, their Figure 1d). This is fast enough to explain the observed latitudinal distribution of HCl.

The combination of enhanced dust and water abundances seems to be the key ingredient for the production of HCl. This may explain why there is no HCl detected in the northern summer season, which, despite the presence of large amounts of water vapor, is relatively dust-free. HCl may still be there, but below the detection limit. Dust optical maps from Montabone et al. (2015, 2020) show a ratio of dust optical depths between perihelion and aphelion of  $\sim 5$  (even in a low dust year), which could be sufficient to either preclude the formation of HCl in northern summer, or to keep it below detectability. However, detection limits reported in Korablev et al. (2021) (0.1–0.2 ppbv) and here (0.3 ppbv) are much lower than simply scaling the observed HCl abundances by a factor of 5. It is possible that a certain threshold of dust abundance is required for the formation of HCl. Another possibility is that the HCl is released from the southern polar cap together with water vapor. But then the question remains why the same process is not occurring at the northern summer cap. If chlorine species are stored in the seasonal caps and released with water vapor during local spring and summer to form HCl, HCl should be strongly present during northern summer, which is not the case.

### 4.2. Sink of HCl

Another important issue arising from the reported observations regards the relatively rapid decay of atmospheric HCl abundances, requiring a strong HCl loss rate (Korablev et al., 2021 and Figure 2 of this article). Indeed, the photochemical lifetime of HCl in the low atmosphere of Mars is estimated to be relatively long (see supporting information, Section S2). It can vary between  $>90$  sols to  $>1,000$  sols below 15 km at the southern and northern mid-latitudes during the northern hemisphere winter solstice (see Figure S2). On the other hand, the production of HCl through the reaction between Cl and HO<sub>2</sub> is much faster (typically on timescales  $<10$  min in the lower atmosphere; see Figure S2). As a result, HCl is expected to be the primary Cl reservoir species in the Martian atmosphere. In this context, the sudden loss of HCl which is observed around  $L_s = 320^\circ$  (Figure 2) cannot be explained by gas-phase photochemistry alone. A sink process, in addition to photochemical loss, has to be invoked. Korablev et al. (2021) suggested that HCl could be removed from the atmosphere by uptake in growing water ice clouds particles and/or surface ice (Kippenberger et al., 2019). The striking similarity between HCl and water vapor vertical profiles shown in this study suggests that the vertical distribution of HCl is controlled by water ice clouds. However, the sublimation of those ice particles would be accompanied by the release of chlorine and thus the formation of atmospheric HCl, and thus it looks problematic to explain the permanent removal of HCl from the atmosphere by the uptake of HCl (reversible or irreversible) by water ice clouds.



Another possibility to remove HCl may be heterogeneous reactions on water ice cloud particles. This process is well known on Earth, where such reactions play a role in the formation of the polar ozone hole:  $\text{ClONO}_2 + \text{HCl} \rightarrow \text{Cl}_2 + \text{HNO}_3$ ,  $\text{N}_2\text{O}_5 + \text{HCl} \rightarrow \text{ClNO}_2 + \text{HNO}_3$ ,  $\text{HOCl} + \text{HCl} \rightarrow \text{Cl}_2 + \text{H}_2\text{O}$  (Solomon, 1999). However, even if those reactions could take place efficiently on Mars, the produced chlorinated compounds would be readily converted to Cl radicals by photolysis (on timescales of a few tens of minutes), and subsequently create HCl by Reaction (1). Chlorine can also be removed by the formation of perchlorates ( $\text{ClO}_4^-$ ). The uptake of ClO on ice particles was experimentally observed to lead to the release of chlorine dioxide (OCIO) upon ice evaporation (McKeachie et al., 2004). Although the enhanced concentrations of OCIO could favor the production of  $\text{HClO}_4$  (Catling et al., 2010; Wilson et al., 2016) and the permanent removal of chlorine from the gas phase, this process should be much too slow to explain the rapid loss of HCl.

It remains difficult to explain the disappearance of HCl after  $L_s = 320^\circ$ . Interestingly, the time of this sudden HCl loss corresponds to the onset of the regional storm (Figure 2) that is annually repeated and recognized as “c” dust storm. This raises the possibility that this dust storm acts as the sink for HCl. In this respect, note that mineral dust was found to be the seat of HCl removal in the Earth’s atmosphere (Sullivan et al., 2007). However, these arguments require robust dust-HCl interactions that are not yet established by laboratory studies, and so need to be addressed in the future.

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#### 5. Conclusions

In this study, we present the full data set of vertically resolved HCl detections derived from the TGO/NOMAD measurements taken between April 2018 and December 2020, which covers southern summer periods in MY 34 and 35. The main results are as follows:

1. We find enhancement of HCl at the southern summer period ( $L_s = 240\text{--}320^\circ$ ) both in MY 34 and 35. It suggests that the global dust storm (which occurred only in MY 34) is not necessary for the formation of HCl.
2. Both in MY 34 and 35, a larger abundance of HCl is observed at the southern high latitudes where water vapor sublimates from the southern polar cap.
3. We find that the vertical distributions of HCl and water vapor are strikingly similar in shape. It suggests that once HCl is introduced in the atmosphere, it follows water vapor including uptake in water ice clouds.
4. We show that the lifetime of HCl in the Mars atmosphere is too long to explain the observed rapid loss of HCl in the Mars atmosphere around  $L_s = 320^\circ$ , which thus requires a strong sink of HCl independent from the photochemical loss.

#### Data Availability Statement

The NOMAD data used in this study are available from ESA’s Planetary Science Archive at <https://archives.esac.esa.int/psa/#!/Table%20View/NOMAD=instrument>.

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